



Graphene based supercapacitors

Paolo Bondavalli
Thales Research and Technology

31 Mars 2016, Nanolnnov Saclay

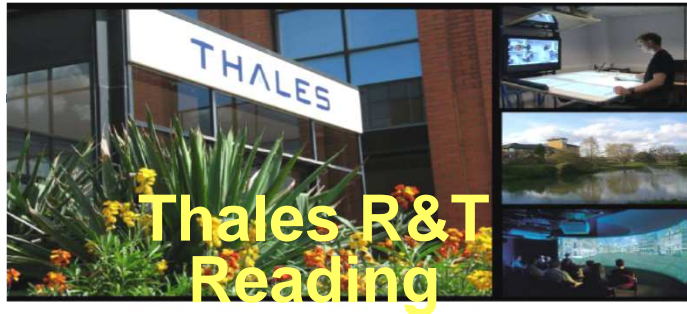
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Mission

- THALES Research & Technologies is a portal for emerging technologies into THALES Group
- Open organisation, co-located close to or within some of the best research campus in our fields, according to the Group worldwide map of locations
 - **France (Palaiseau) : 350 p + 70 PhD + 80 CNRS-Universities**
 - Ecole Polytechnique – Plateau de Saclay
 - **UK (Reading) : 130 p**
 - University of Surrey
 - **Netherlands (Delft) : 15 p**
 - Technological University of Delft - University of Twente
 - **Singapore : 15 p**
 - Nanyang Technical University

The actors of research

THE JOINT LABS

TRT

INNOVATION PLATFORMS

Work on concrete cases of technology implementation in the products of the group.

DIVISIONS

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ACADEMICS JOINT LABS

Those common labs have common personnel, common equipment and shared research agenda

ACADEMICS
CNRS-Ecole
Polytechnique...

TRT

INDUSTRIAL JOINT LABS

Those common labs have common personnel, common equipment and shared research agenda

INDUSTRIALS
ALCATEL



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What's graphene? What are carbon nanotubes?

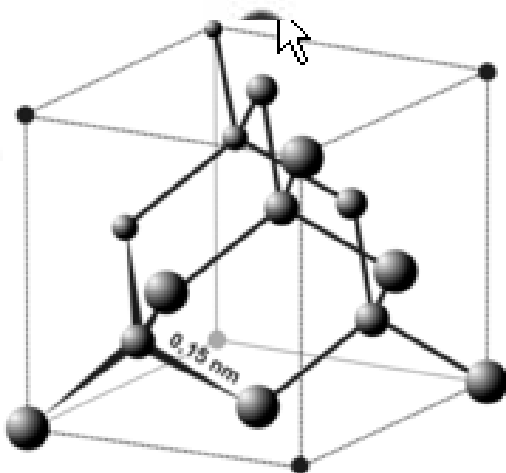


Allotropic forms of Carbon (until 80s)

These different ways of rearrange themselves, change the properties of the final “macromaterials”. For example graphite is black and quite soft, inversely diamond is transparent and one of the hardest materials discovered up to now.

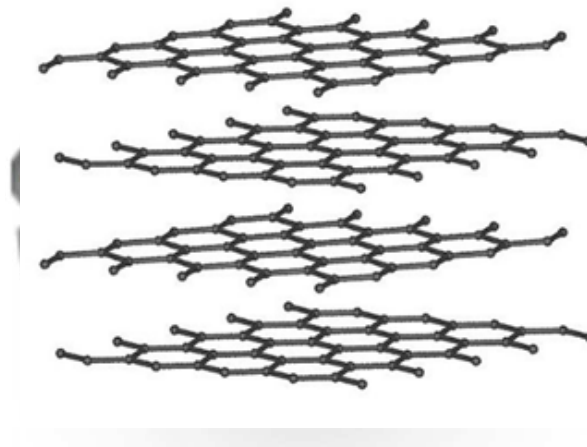


Diamond



4 Bondings

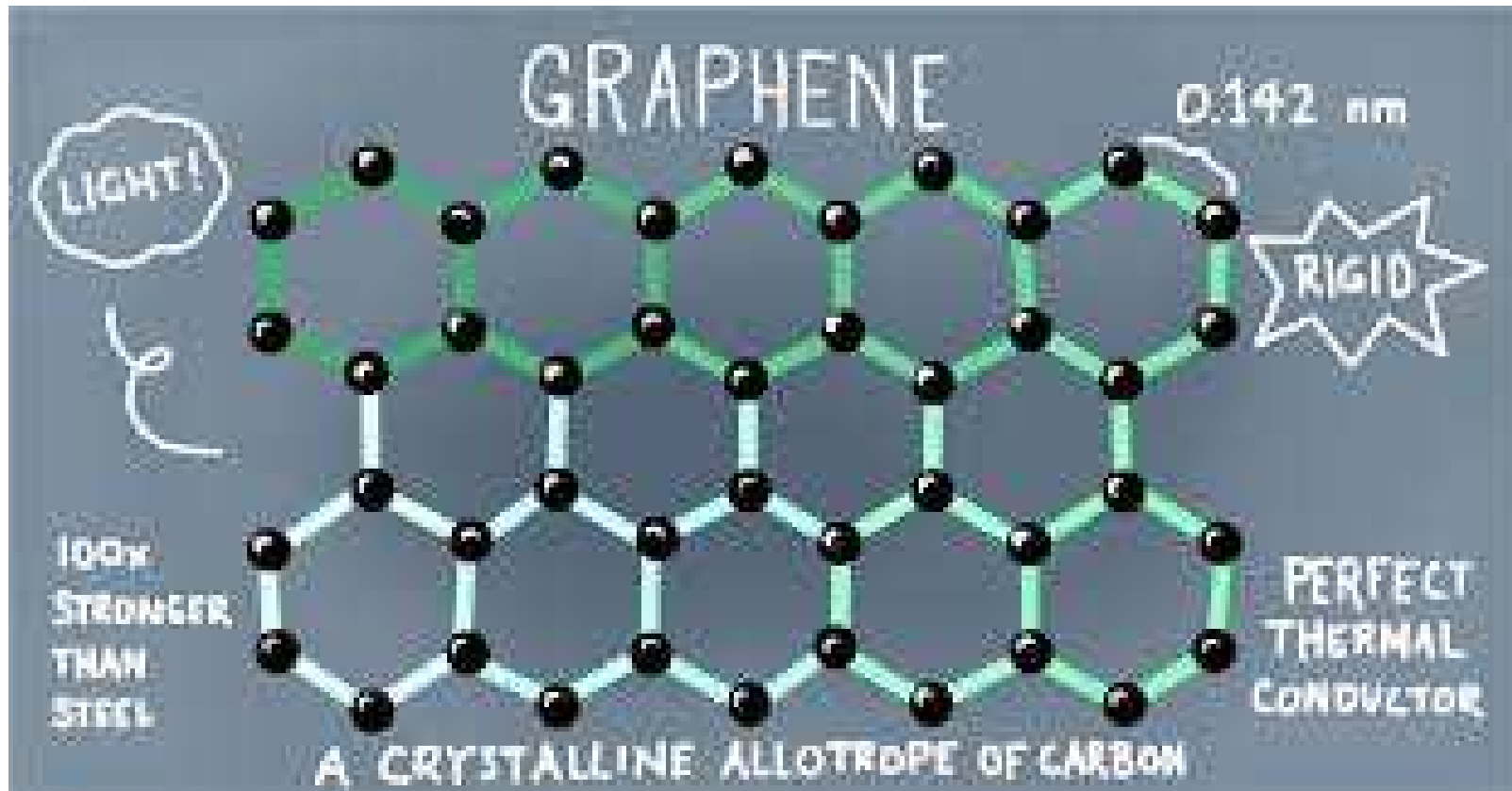
Graphite



3 Bondings

After 80s....

What's graphene ?



The Nobel Prize in Physics 2010



Photo: U. Montan

Andre Geim

Prize share: 1/2



Photo: U. Montan

Konstantin Novoselov

Prize share: 1/2

1946 Theoretical Prediction (P. R. Wallace, Un. Ontario)

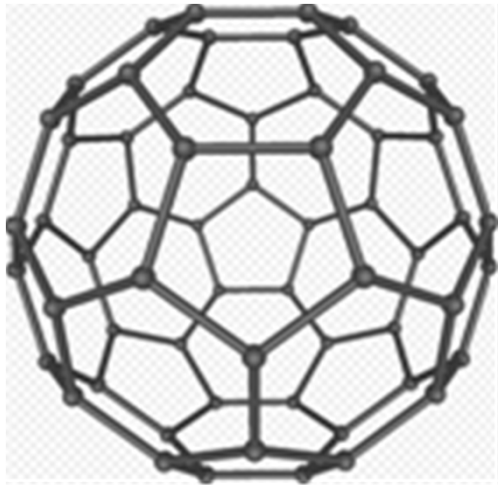
**1961 First observation of "2D Graphite Lamellae"
(Hanns-Peter Boehm, Ludwig-Maximilians-Universität in
Munich)**

The Nobel Prize in Physics 2010 was awarded jointly to Andre Geim
and Konstantin Novoselov *"for groundbreaking experiments
regarding the two-dimensional material graphene"*

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Fullerene : beautiful...but no real application...



Buckminsterfullerene C₆₀, where 60 is the number of carbon atoms, is the most known form of the fullerene family.

It was first generated in 1985 by Harold Kroto, James R. Heath, Sean O'Brien, Robert Curl, and Richard Smalley at Rice University

1996 Nobel Prize in Chemistry



Richard Buckminster Fuller

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HALES

The Nobel Prize in Chemistry 1996



Robert F. Curl Jr.

Prize share: 1/3



Sir Harold W. Kroto

Prize share: 1/3



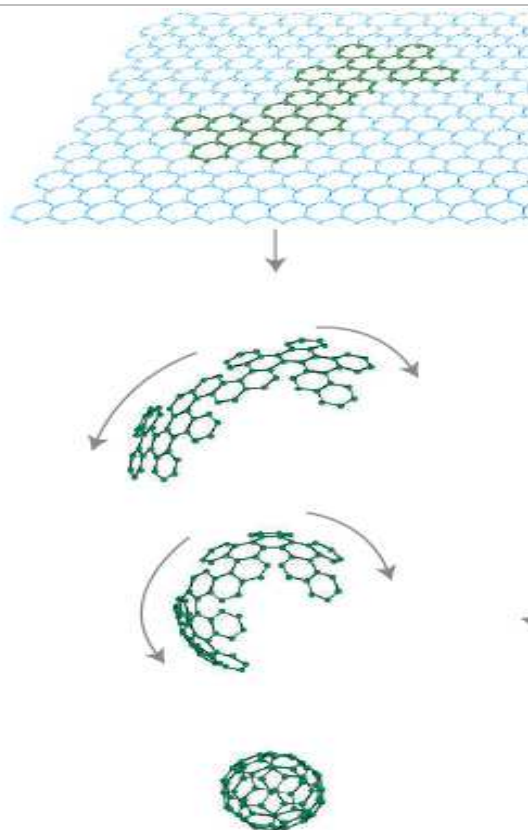
Richard E. Smalley

Prize share: 1/3

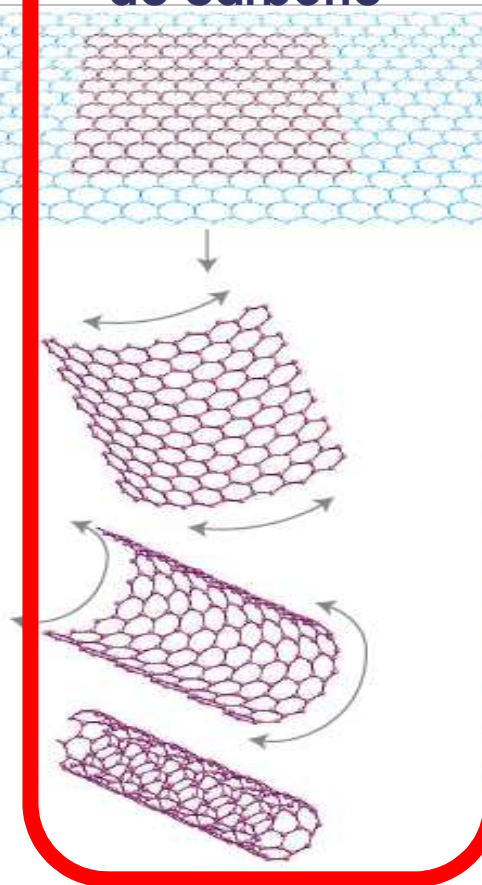
The Nobel Prize in Chemistry 1996 was awarded jointly to Robert F. Curl Jr., Sir Harold W. Kroto and Richard E. Smalley *"for their discovery of fullerenes"*.

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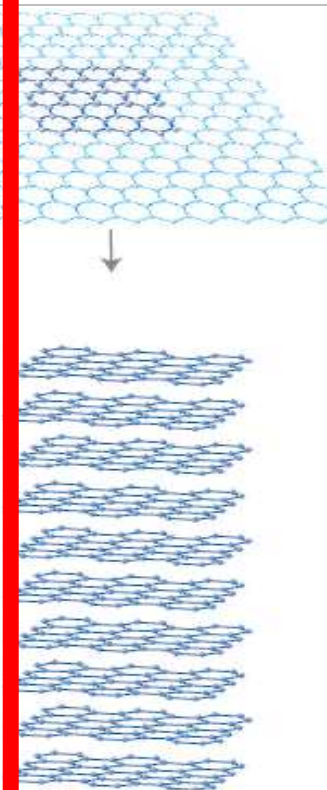
Fullerene/Buckyballs



Nanotubes de carbone



Graphite

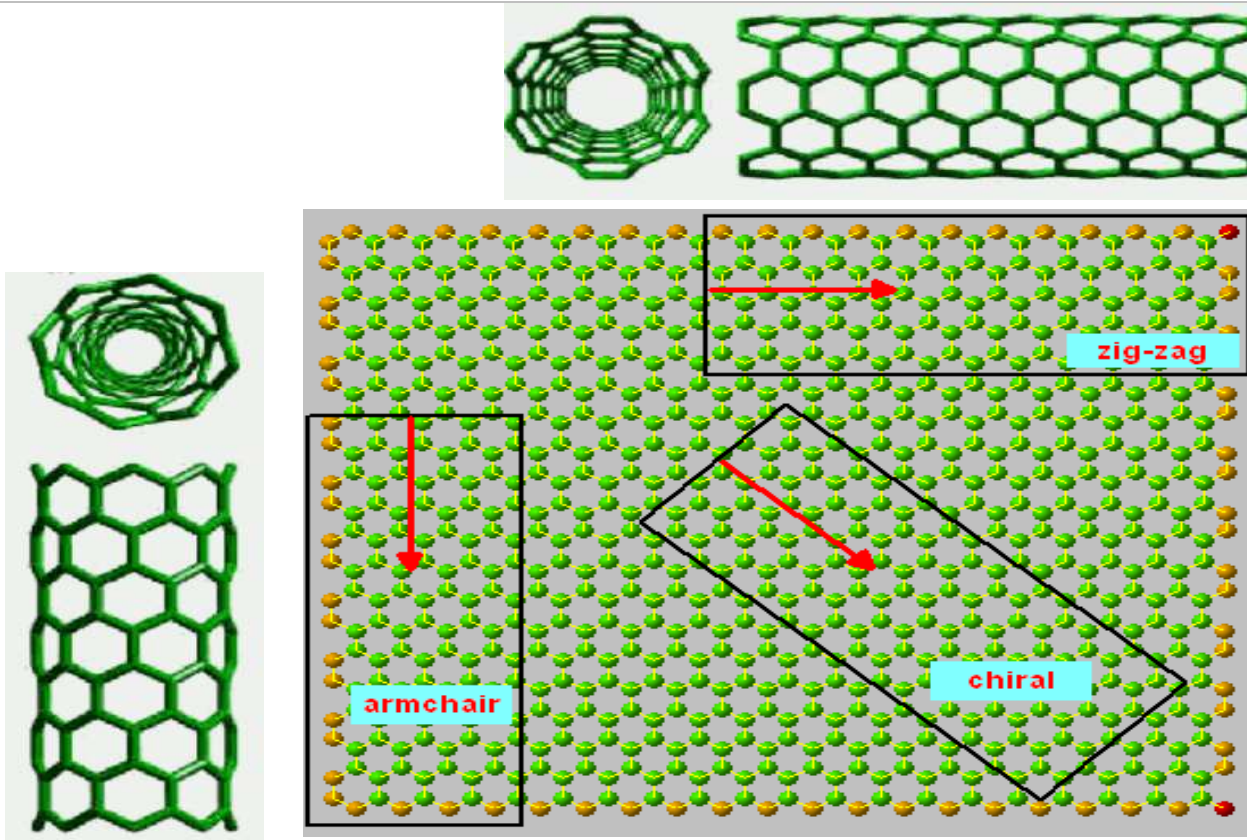


1991 Carbon Nanotubes Discovery

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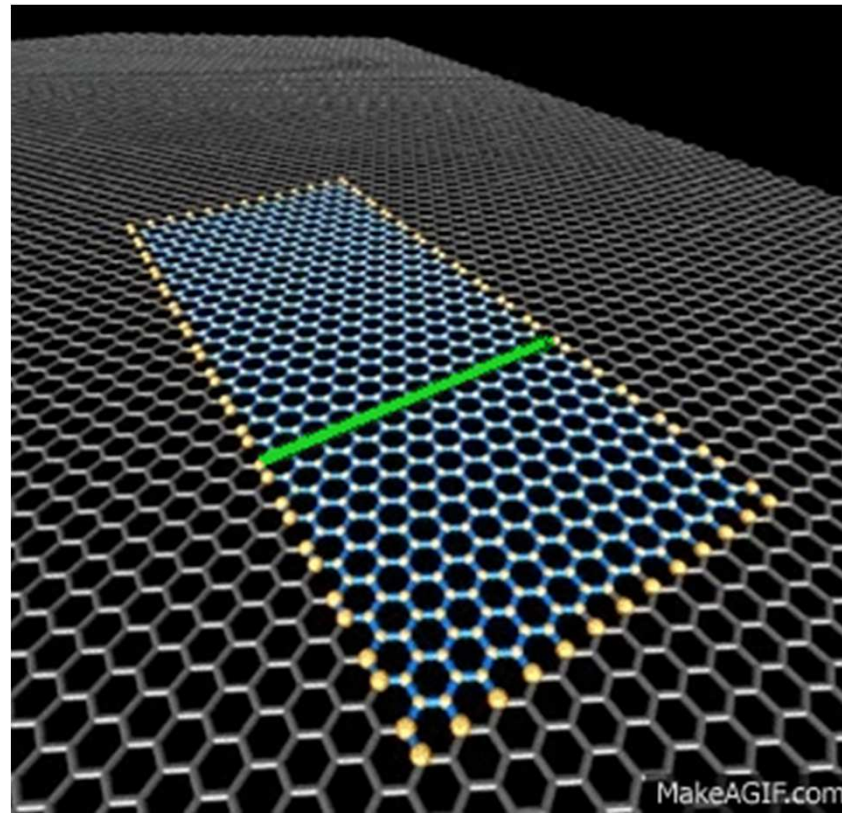
Carbon nanotubes properties : how to roll up graphene?



Graphene

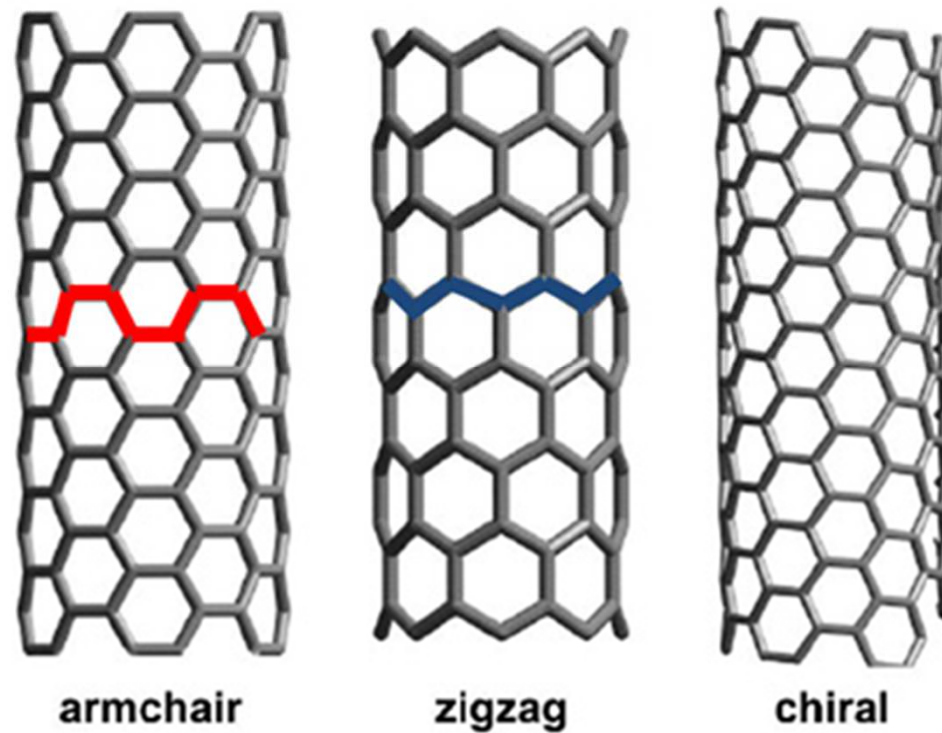
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Rolling up Graphene to Obtain Carbon Nanotubes

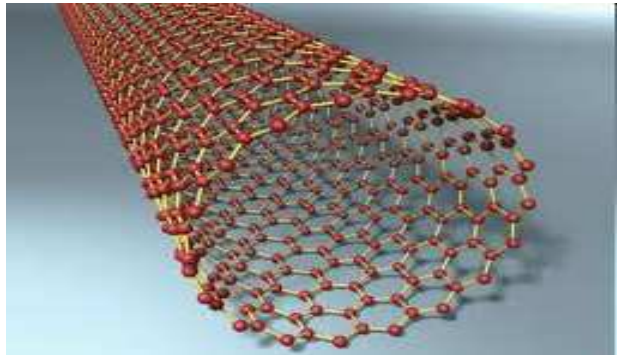


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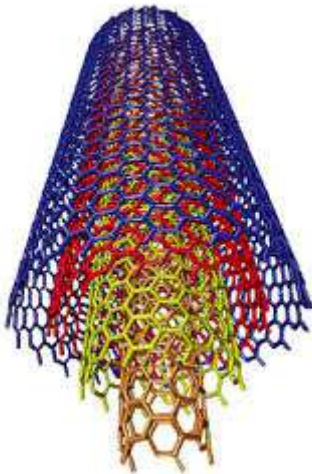
Carbon nanotubes properties : how to roll up graphene?



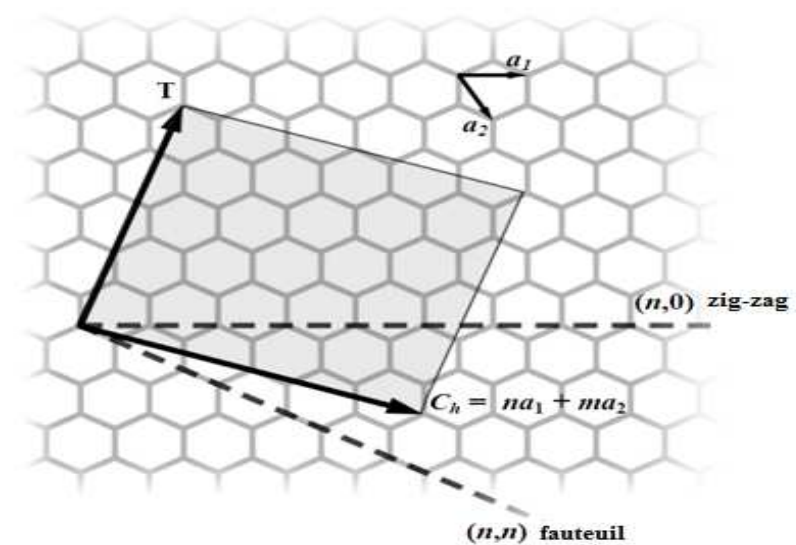
MultiWalled Carbon Nanotubes, SingleWalled Carbon Nanotubes



SWCNT



MWCNT



$$\underline{C}_h = n.\underline{a}_1 + m.\underline{a}_2$$

Saito Law :

$$n-m \neq 3p \text{ (p entier)}$$

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Sumio Iijima

Senior Research Fellow, NEC Corporation

新素材の力で
日本のものを
加速したい。



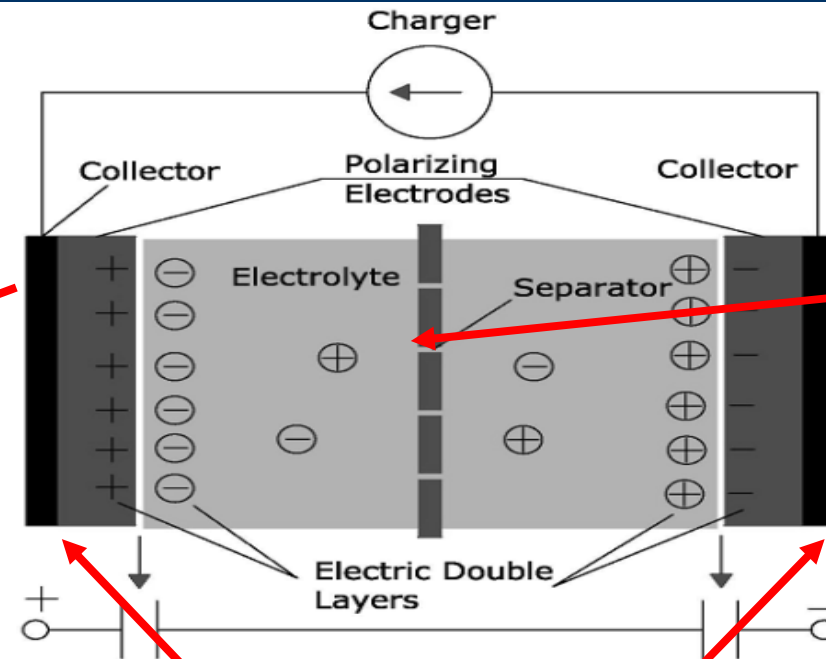
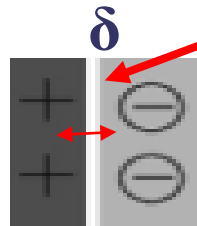
NOBEL PRIZE

What's a supercapacitor?

Supercapacitor is the name done by NEC in 1971
Technically is defined Electrical Double Layer Capacitor (EDLC)

$$C = (\epsilon/\delta)A$$
$$E = (1/2)CV^2$$
$$P = V^2/(4R)$$

Helmutz's model

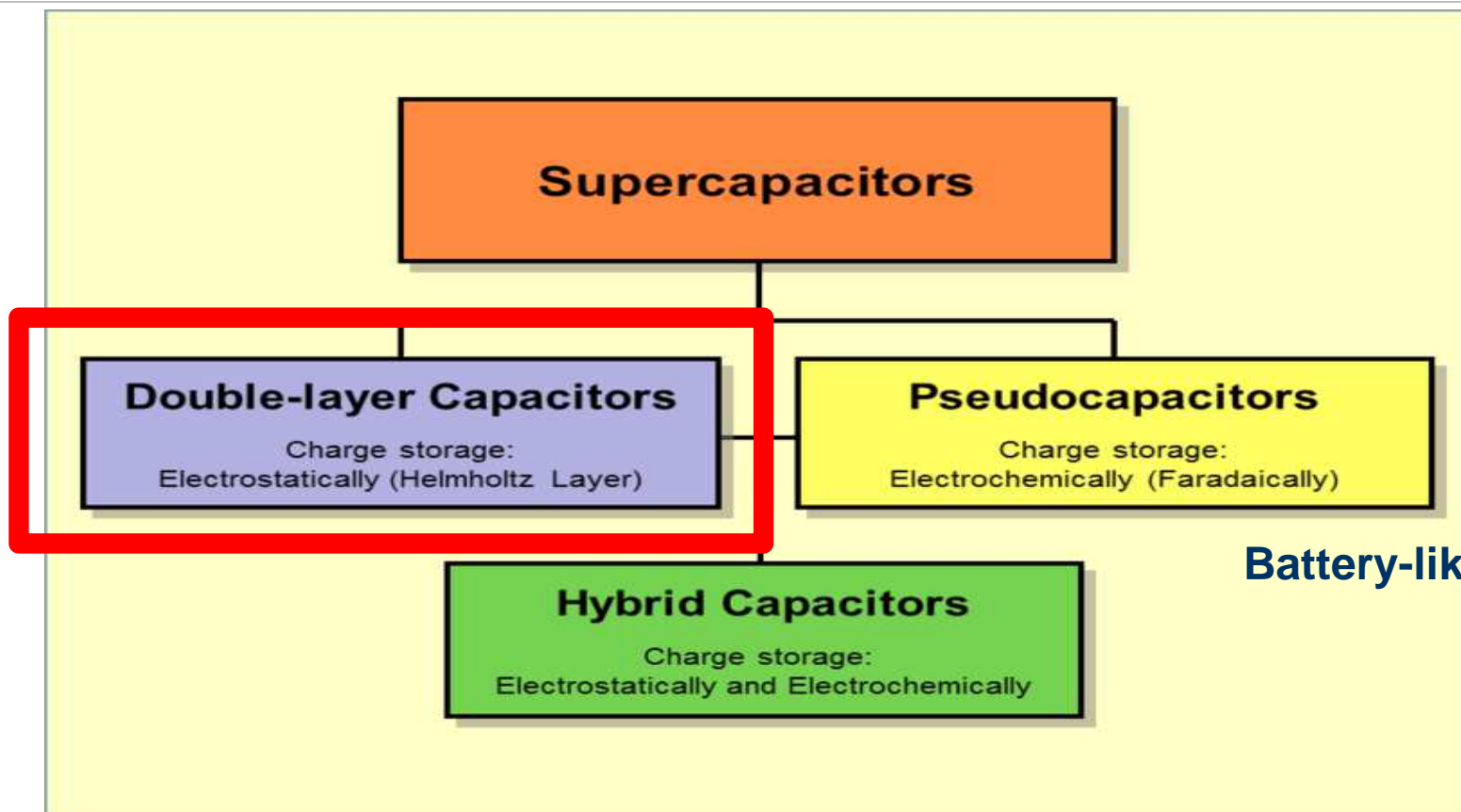


separator

Only non faradic reactions

Two electrodes

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Battery-like devices

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Advantages

- Very high rates of charge and discharge
- Higher life cycle (>500000, rechargeable batteries can attain 10000)
- Good reversibility
- Low toxicity of material used
- High cycle efficiency
- Low internal resistance (Higher output power)
- Extremely low heating levels



Drawbacks

- Low amount of energy stored (3-5 Wh/Kg vs 30-40 Wh/Kg for batteries)
- It requires sophisticated control and switching equipment (from batteries to supercaps)



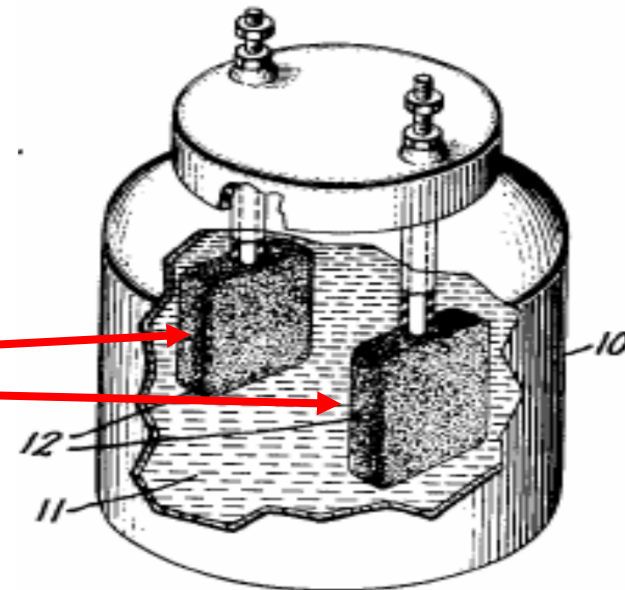
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A short history

1957 : the first patent on EDLC (Electrical Double Layer Capacitance) General Electric : Electrolytic capacitor with porous carbon electrodes

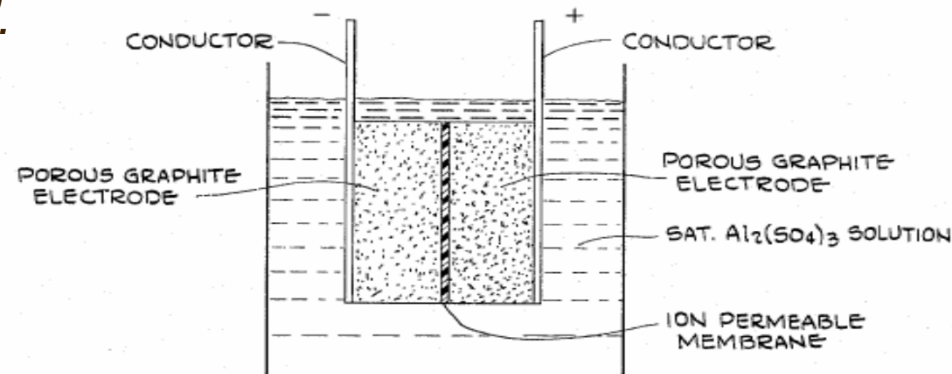
2,800,616
LOW VOLTAGE ELECTROLYTIC CAPACITOR
Howard I. Becker, Viscers Ferry, N. Y., assignor to General Electric Company, a corporation of New York
Application April 14, 1954, Serial No. 423,042
4 Claims. (Cl. 317—230)

Two porous graphite electrodes



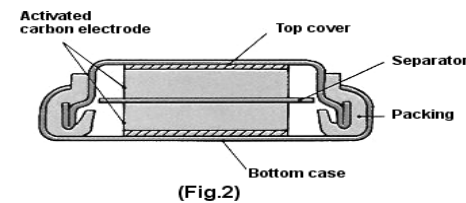
1966 : SOHIO (Standard Oil company, OHio)

At this time SOHIO acknowledged that “*the double-layer at the interface behaves like a capacitor of relatively high specific capacity*”.



1971 : NEC : first commercial device called « supercapacitor » based on technology suggested by SOHIO: Memory backup applications (low voltage, high internal resistance)

1978 : Panasonic : « gold capacitor » : back-up energy source for microprocessors and solar batteries (power failures include video recorders, DVD players, fax machines, telephones, digital still cameras, mobile phones, audio stereo systems, etc.)



1982 : Pinnacle Research Institute : « PRI ultracapacitor » (first time that the term « ultracapacitor » is used) incorporated metal-oxide electrodes

Applications : laser weaponry and missile guidance (military applications)

Some visible supercap applications



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Activated carbon: parameters

Main parameters

- Surface (energy)
- High breakdown voltage (energy)
- Pore size (to exploit surface completely and to promote easy ion diffusion)

Activated Carbon

- Large surfaces ($3000\text{m}^2/\text{g}$)
- Low-cost material

The main issue :

- Very bad mesoporous distribution!!!
2/3 of the pore size are smaller than 2 nm and so are unpercolated)

Non-faradic carbon nanotubes based supercapacitors : state of the art, **P.Bondavalli**, et al. , Eur. Phys. J. Appl. Phys. 60,10401, 2012

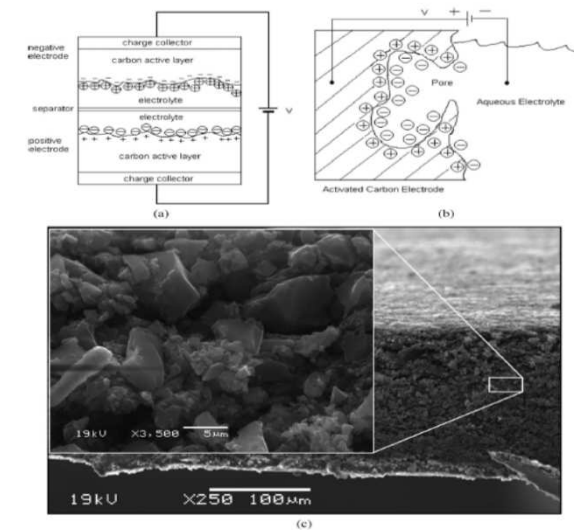
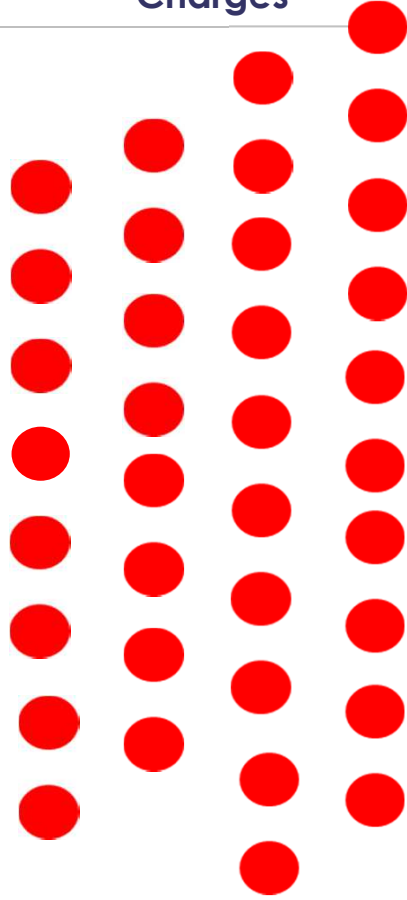


Fig. 1. (a) Schematic of an activated carbon-based EDLC. (b) Representation of pore in carbon electrode active layer. (c) Electron micrograph of activated carbon electrode.

Charges



Electrode

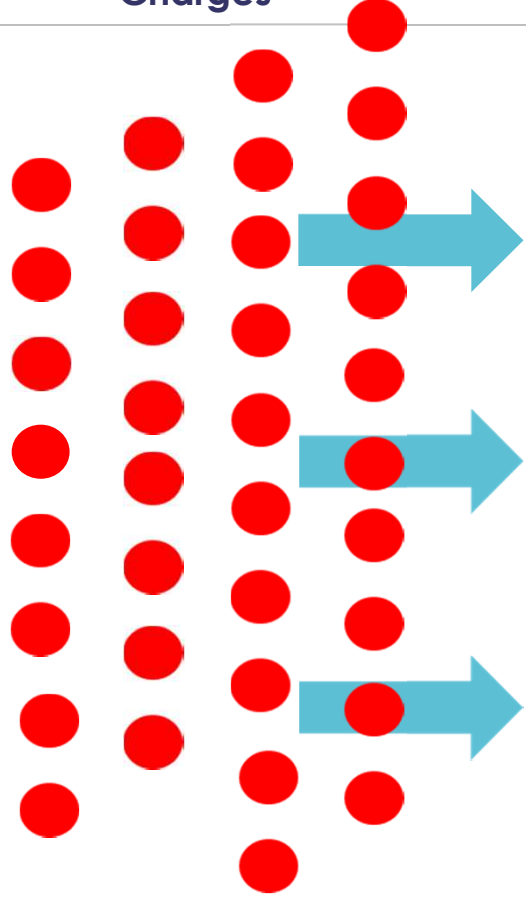


Collector



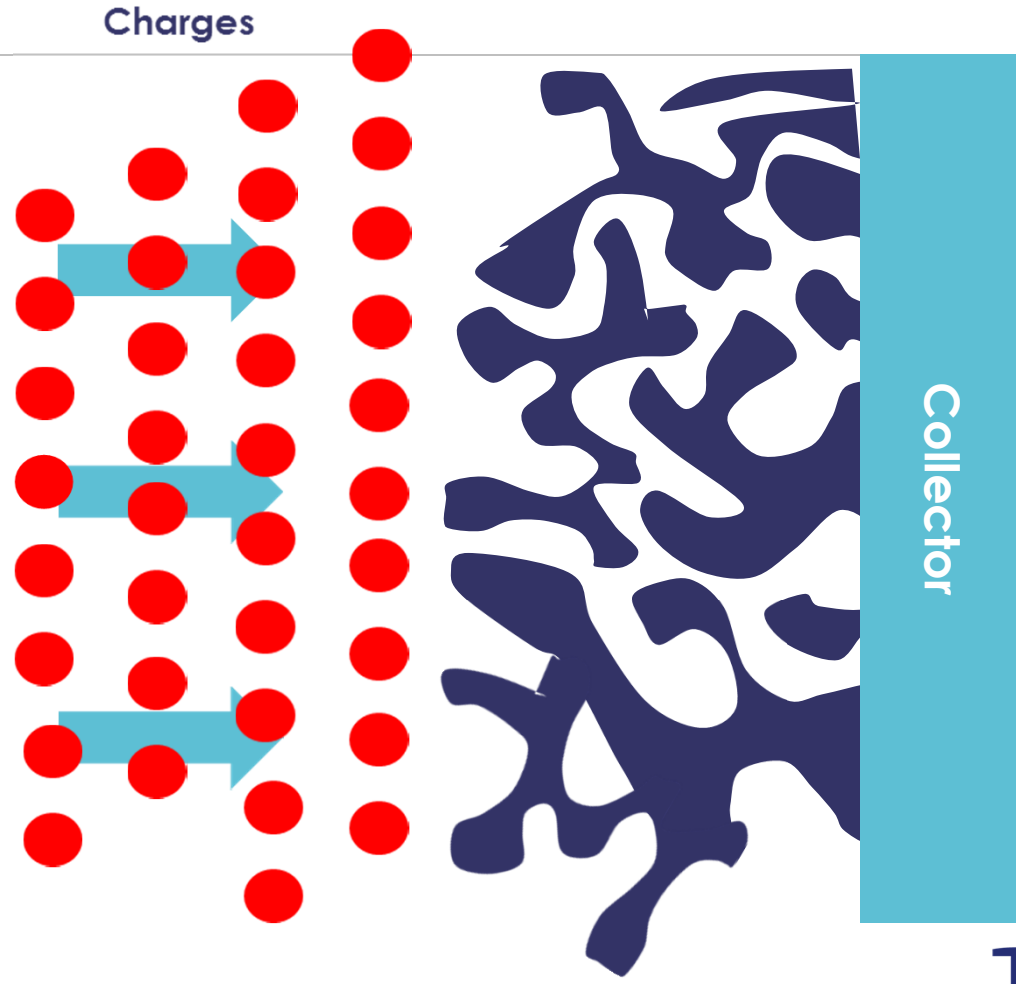
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Charges



Collector

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2/3 of the surface is not exploited

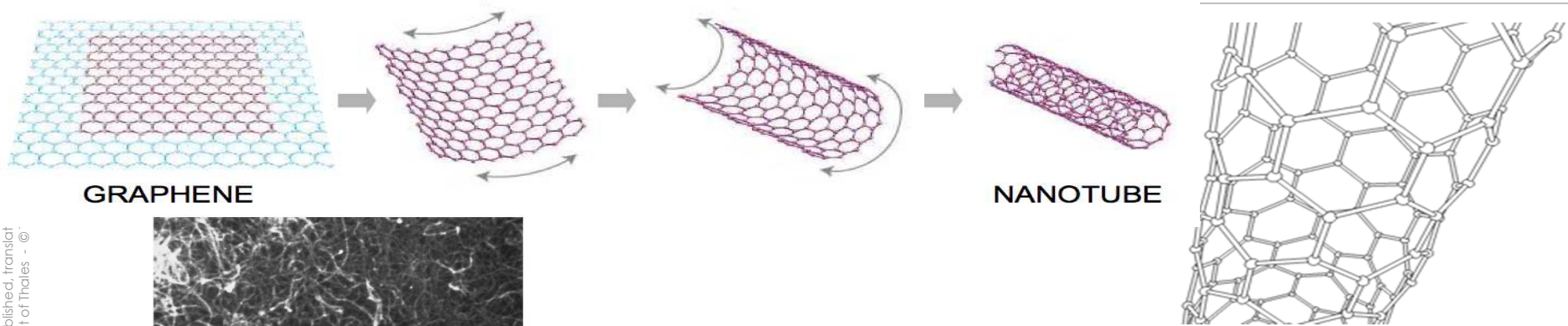


**PORE SIZE IS NOT OPTIMIZED AND
SURFACE IS NOT ADEQUATELY EXPLOITED**



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What are carbon nanotubes and why carbon nanotubes for Supercaps?

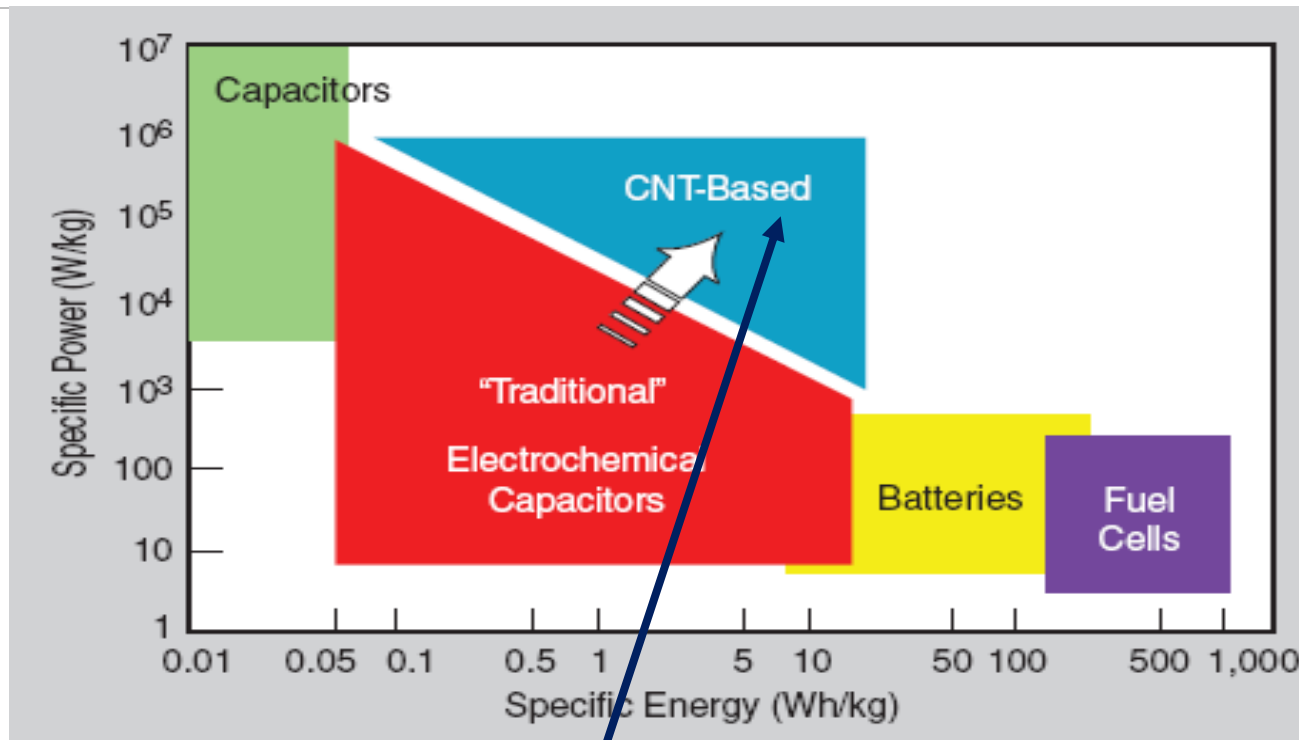


Niu, C.; Sichel, E.K.; Hoch, R.; Moy, D.;
Tennent, H., *Appl. Phys. Lett.* 1997, 70(11)
1480-1482 Hyperonc Inc.

- Randomly entangled nanotubes for electrodes can be fabricated easily
- Highly surface specific surface area ($300\text{m}^2/\text{Kg}$)
- High mesoporous distribution (2-5nm) and so electrolyte accessibility
- Low resistivity (they can be used as electrode and collector)
- We can fabricate electrodes without binder (higher breakdown voltage)
- Total weight is very low (enhancement energy and power density)
- High stability (long life-time)

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CNTs (and Graphene related materials) based Supercapacitors

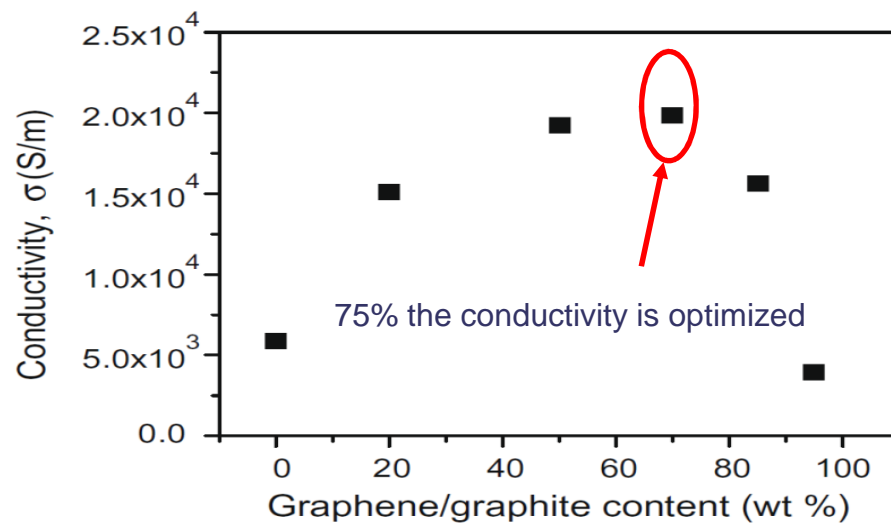


Considering that supercapacitors bridge the gap of capacitors and batteries performances we have to attain performances in this zone

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Why to use Graphene related materials and CNTs mixings?

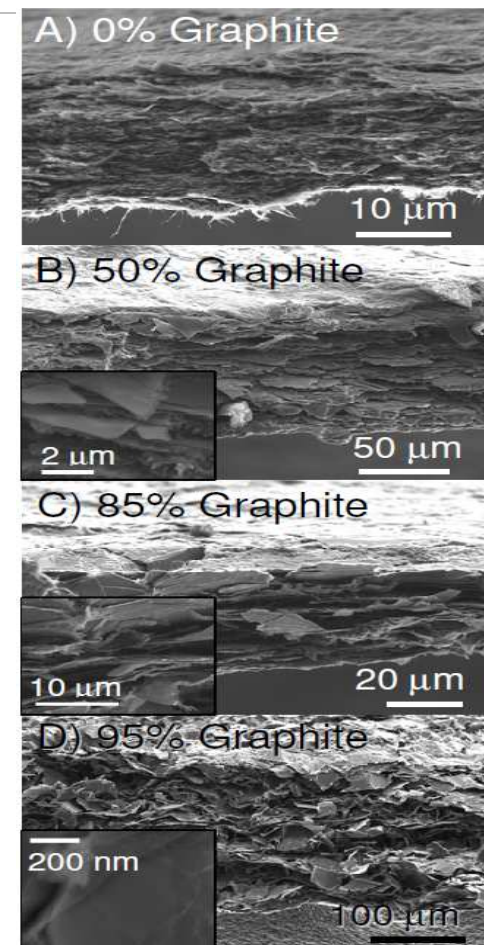
● CNT/graphene/graphite composite



U. Khan, J. N. Coleman et al Carbon (2010)

Resistance is reduced by a factor of 4 compared to bare CNTs layers

Can we improve the Power output ($P \propto 1/R$)?



Why to use Graphene related materials and CNTs mixings?

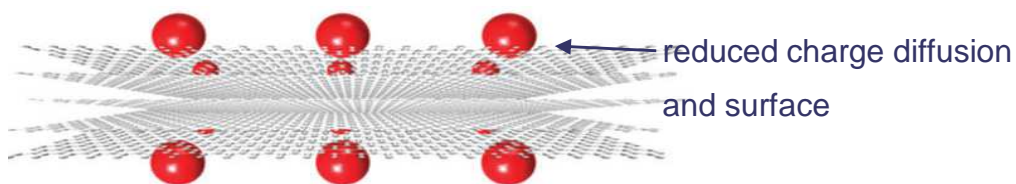
Cite this: *Phys. Chem. Chem. Phys.*, 2011, **13**, 17615–17624

www.rsc.org/pccp

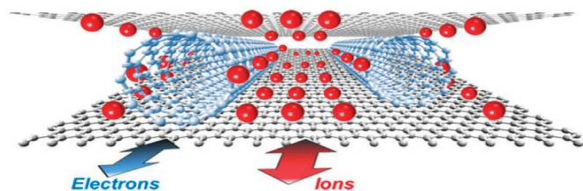
PAPER

Graphene and carbon nanotube composite electrodes for supercapacitors with ultra-high energy density

Qian Cheng,^{ab} Jie Tang,^{ab} Jun Ma,^a Han Zhang,^a Norio Shinya^a and Lu-Chang Qin^a



Pristine
graphene/graphite



Graphene/graphite/CNTs
mixing

- CNTs prevent restacking (higher surface, higher energy stored)
- CNTs/graphite/graphene improve conduction (higher power delivered)
- CNTs prevent the disintegration of the composite

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CNTs



Separated weighing



Separated dispersion
(solvent = NMP)



Dilution to get $C_{\text{solide}} = 0,5\text{g/l}$

Our Approach

Graphite flakes



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Initial sonication

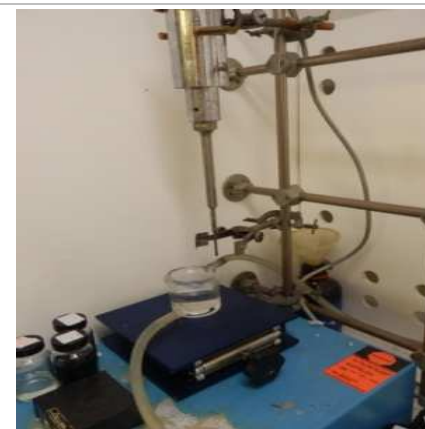
- CNT : 10' high power
- Graphite : 18h low power



Centrifugation 10 minutes x2



OUR APPROACH



OUR APPROACH

Final sonication of the mixture :
18h low power

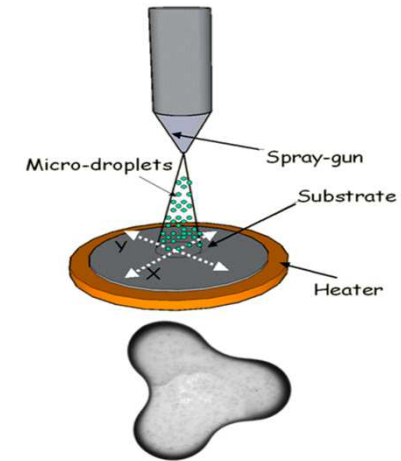
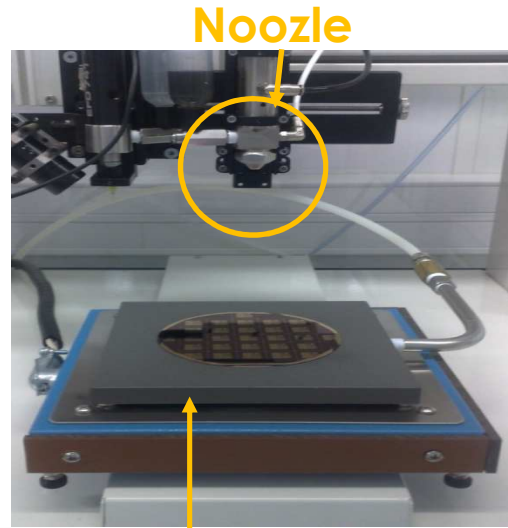
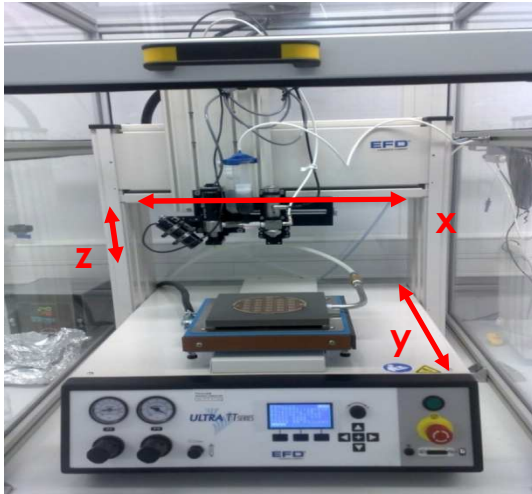


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Dynamic spray-gun deposition method

Deposition method

- Excellent reproducibility
- Versatile, easily scalable for large-area applications
- Extremely uniform deposition with no “coffee-ring” effect



Process patented

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Electrode design and cell fabrication

Supercapacitor electrode based on mixtures of graphite and carbon nanotubes deposited using a new dynamic air-brush deposition technique,
P Bondavalli, JECS 160 (4) A1-A6, 2013

Air-brush deposition

Gun spraying

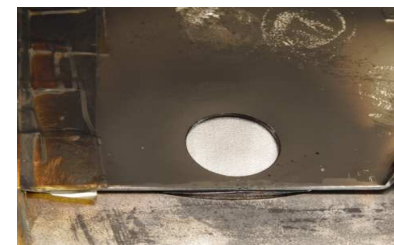
Masking

Several samples fabricated at the same time

Flexible electrodes

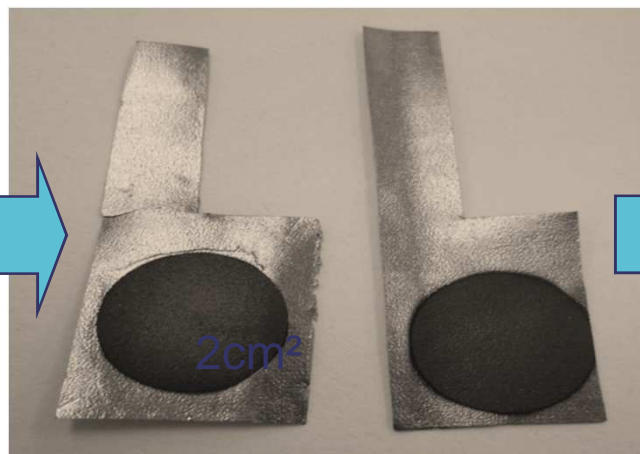
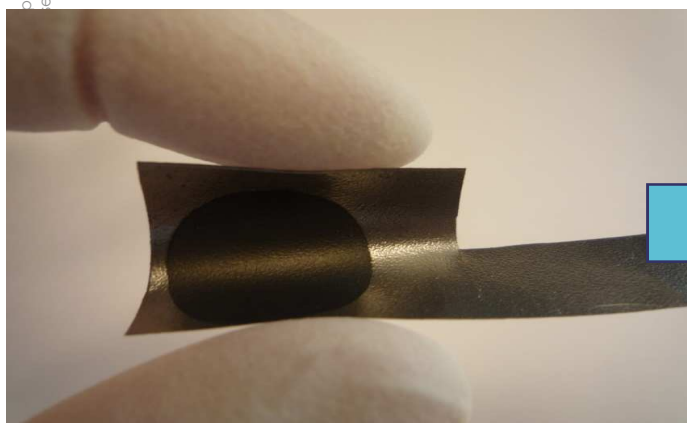


Panasonic
Graphite bucky paper



Electrode design

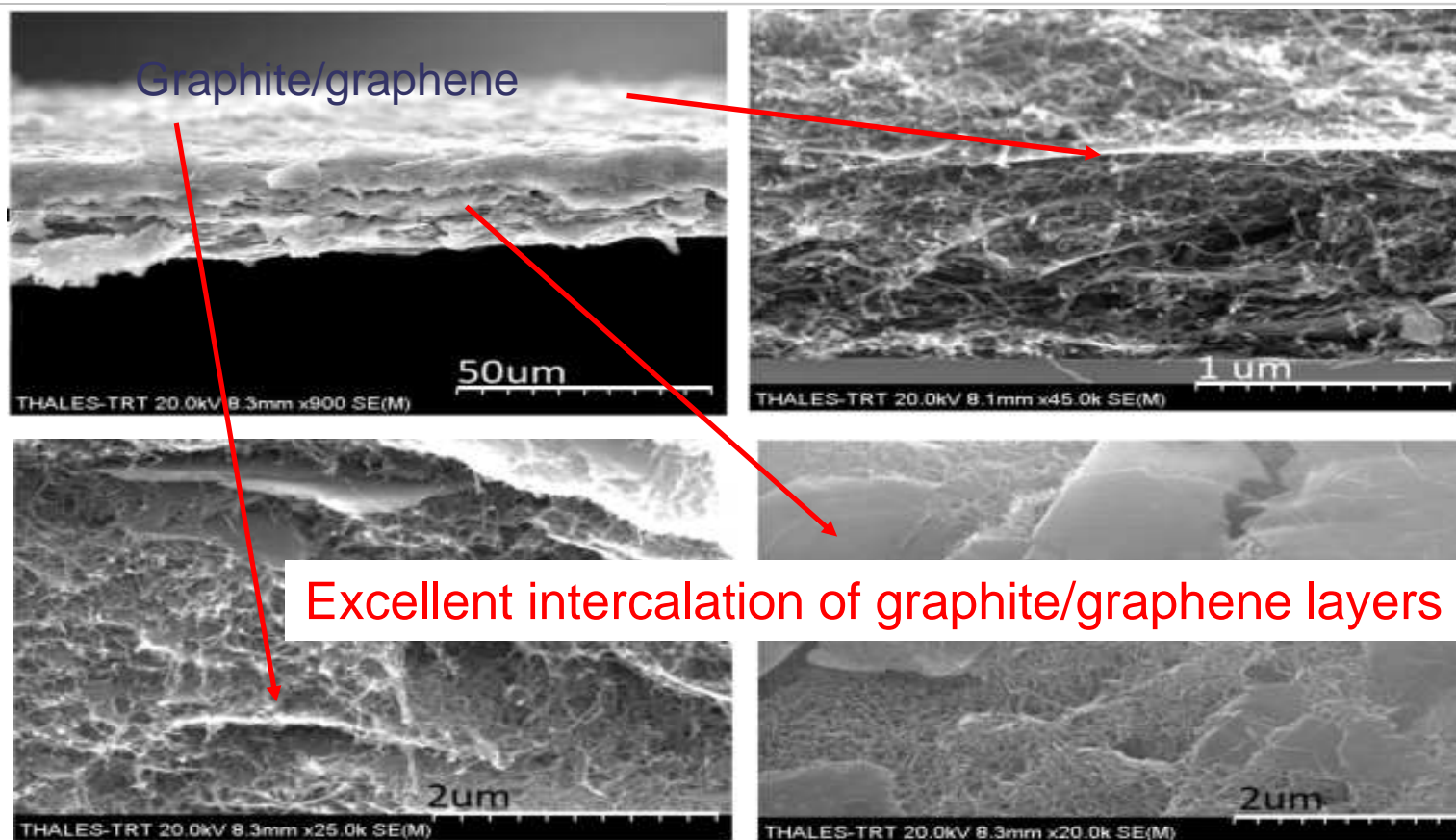
Supercapacitor Cell



This document is a part of

Sample Morphology (cross section)

Supercapacitor electrode based on mixtures of graphite and carbon nanotubes deposited using a new dynamic air-brush deposition technique,
P Bondavalli, JECS 160 (4) A1-A6, 2013



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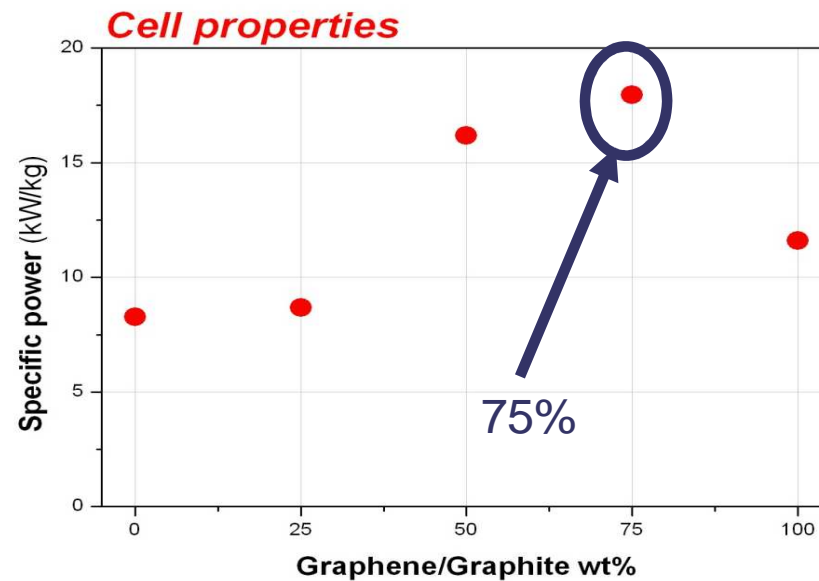
Results : Energy and Power as a function of the concentration

Sample characteristics :

- weight = 1.8mg
- surface = 2cm² (circular design)
- thickness ~ 20µm

A - Influence of the CNT concentration (Electrodes)

- Energy max. ~4,5Wh/kg for 75wt%CNT
- Power max. ~15 kW/kg for 25wt%CNT (enhancement of 2,5)



Supercapacitor electrode based on mixtures of graphite and carbon nanotubes deposited using a new dynamic air-brush deposition technique,
P Bondavalli, JECS 160 (4) A1-A6, 2013

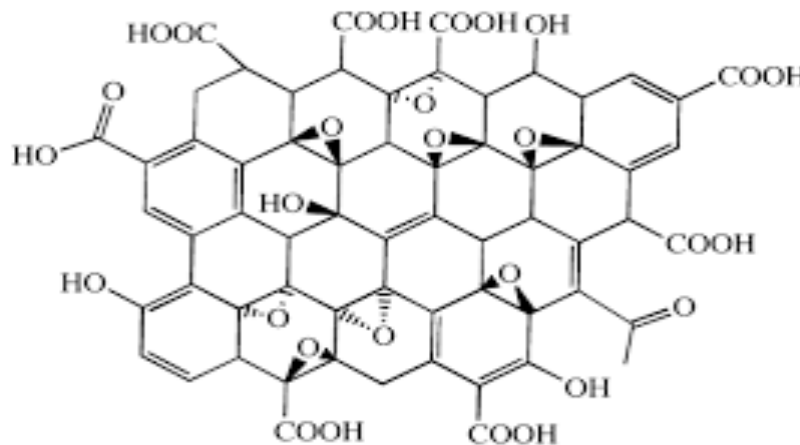
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Last measurements : new option for green suspensions using GO

Mixing of Graphene Oxide and Oxydised Carbon Nanotubes in water

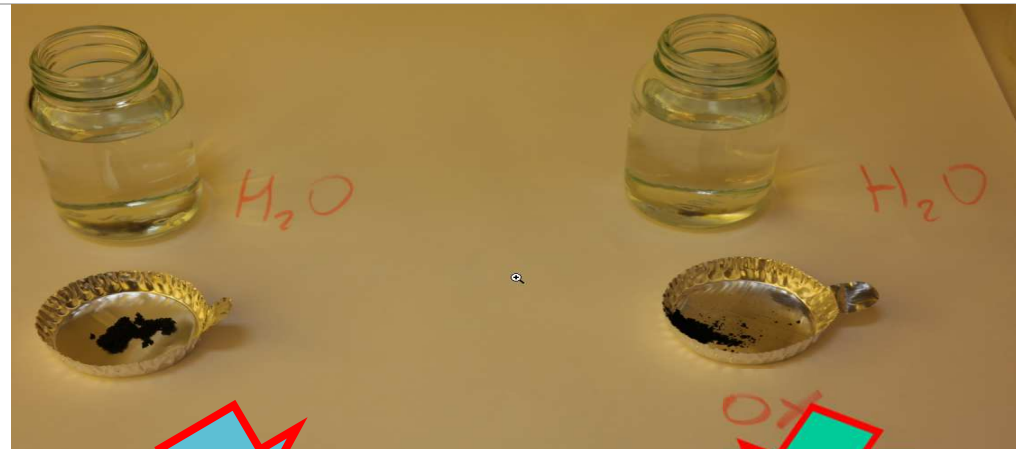
Advantages

- Aqueous based suspensions
- Very stable suspensions
- Low temperature process (120°C)



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Oxidised CNTs can be put into water based suspensions very easily



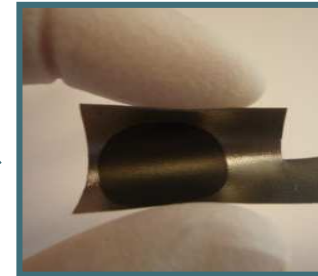
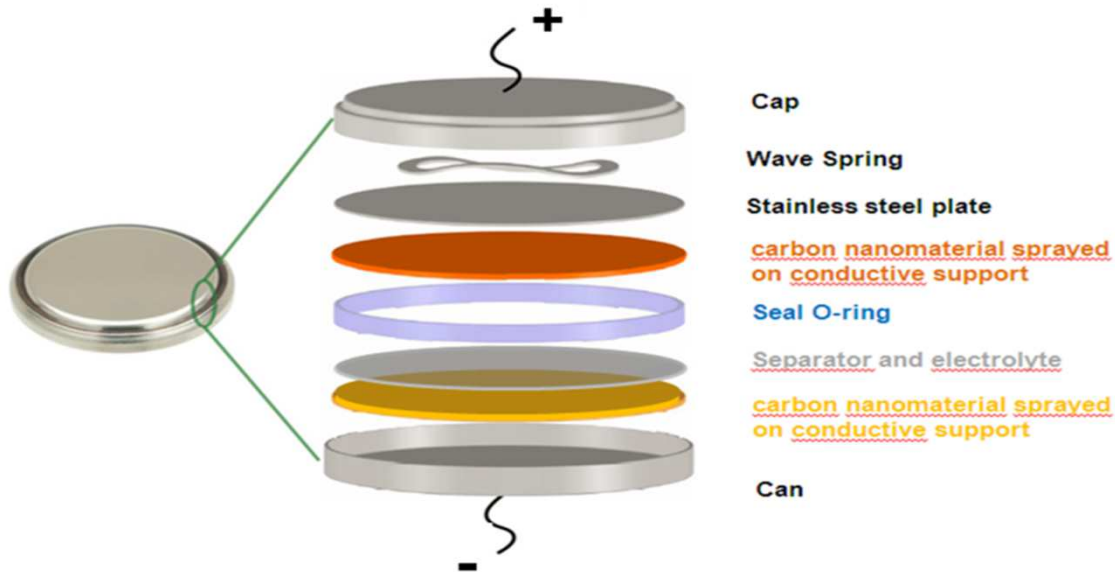
Non-oxidised CNTs



Oxidised CNTs

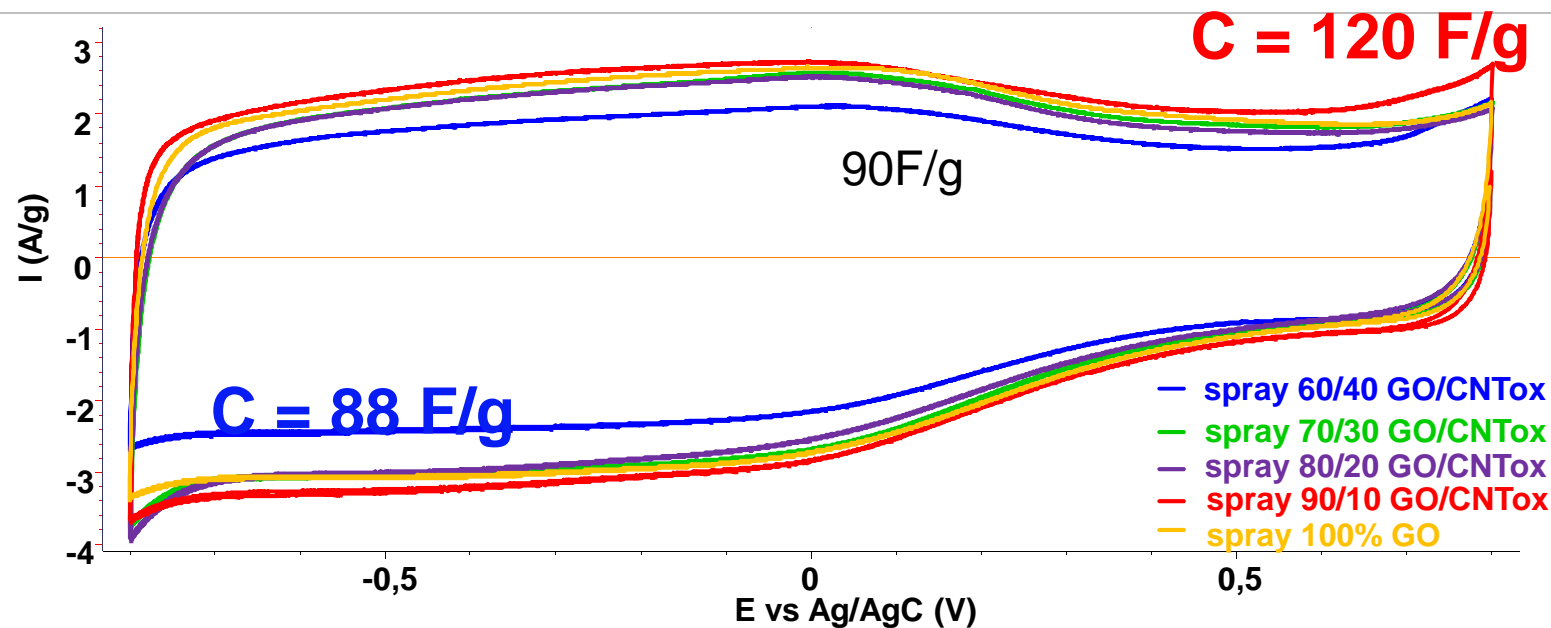


New packaged prototypes



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Performances for different GO concentrations

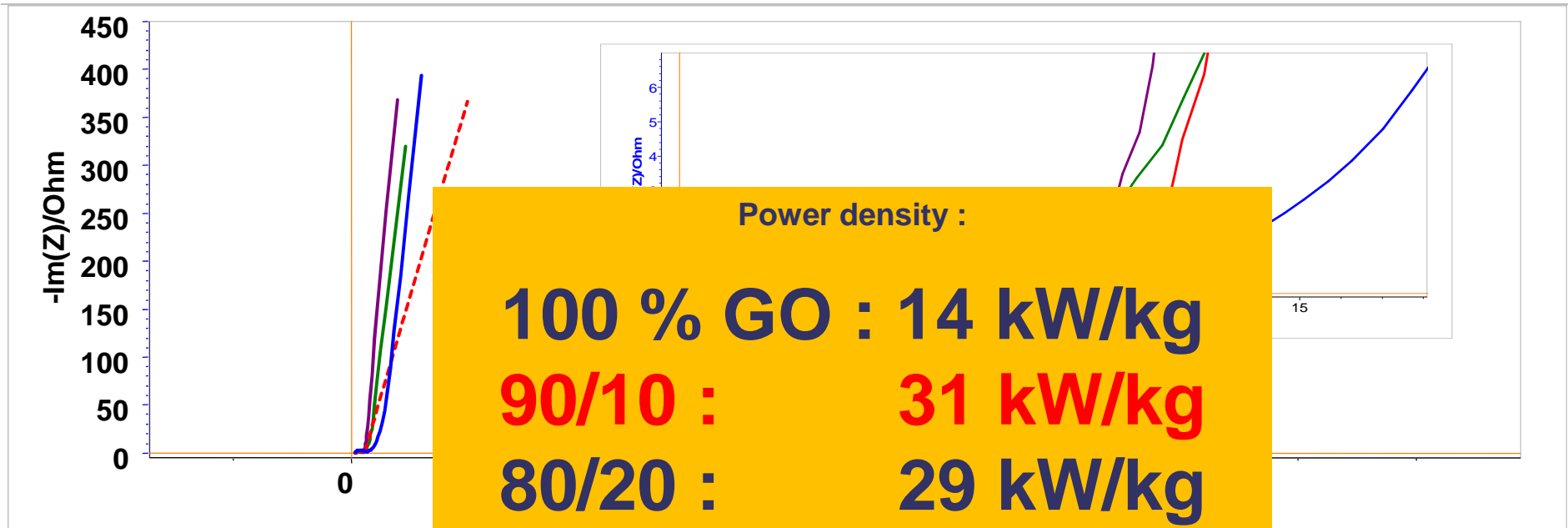


The surface allows to know the capacity of the materials to store energy

Graphene based electrodes for high performances supercapacitors, **P.Bondavalli**, G.Pognon, Proceeding of IEEE NANO 2015, 17-20 July, Rome

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Power density

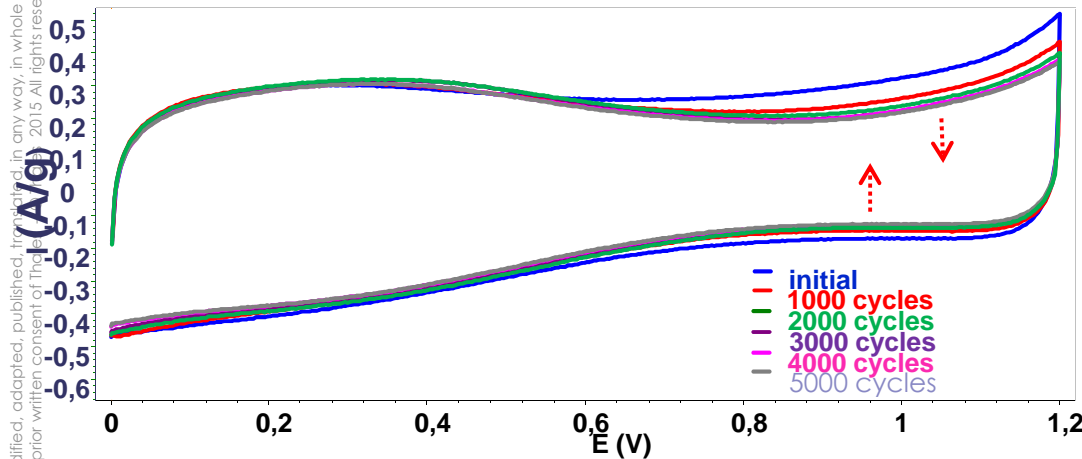


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Graphene from IIT : Galvanostatic charge/discharge experiment

Very good stability (same capacitance that using GO)

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two-electrodes configuration

→ complete system

$m = 1,15 \text{ mg (x2)}$

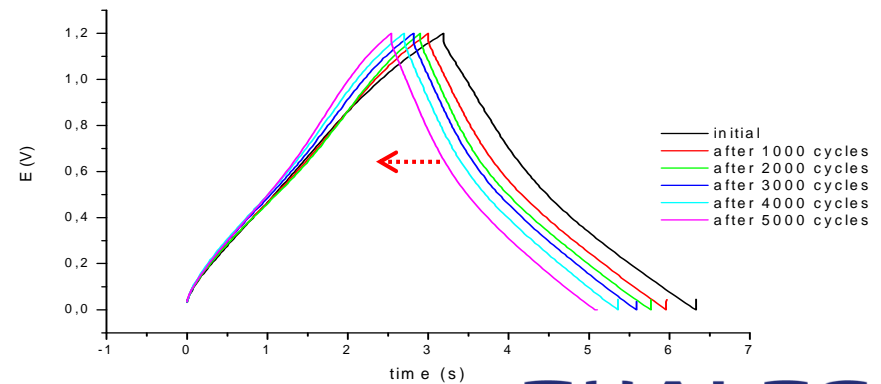
5000 cycles

$I = 10 \text{ mA}$

Graphene exfoliated by IIT :

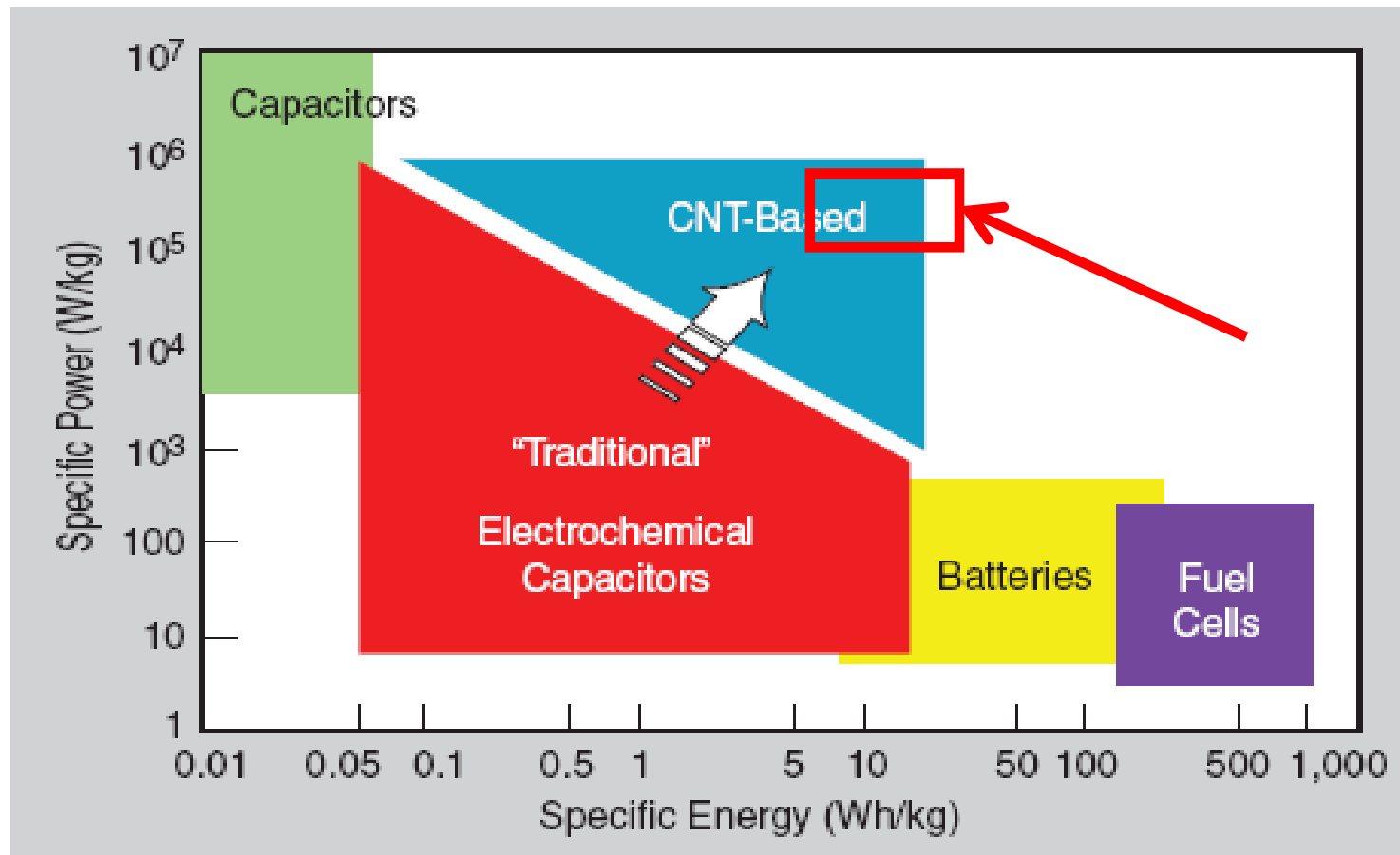
- Loss essentially during the 1000 first cycles

$P = 92,3 \text{ kW/kg}$



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Conclusions and perspectives



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Contracts and Objectives

GRAPHENE FLAGSHIP

➤ Results published in « 2D Materials » and presented at IEEE NANO 2015

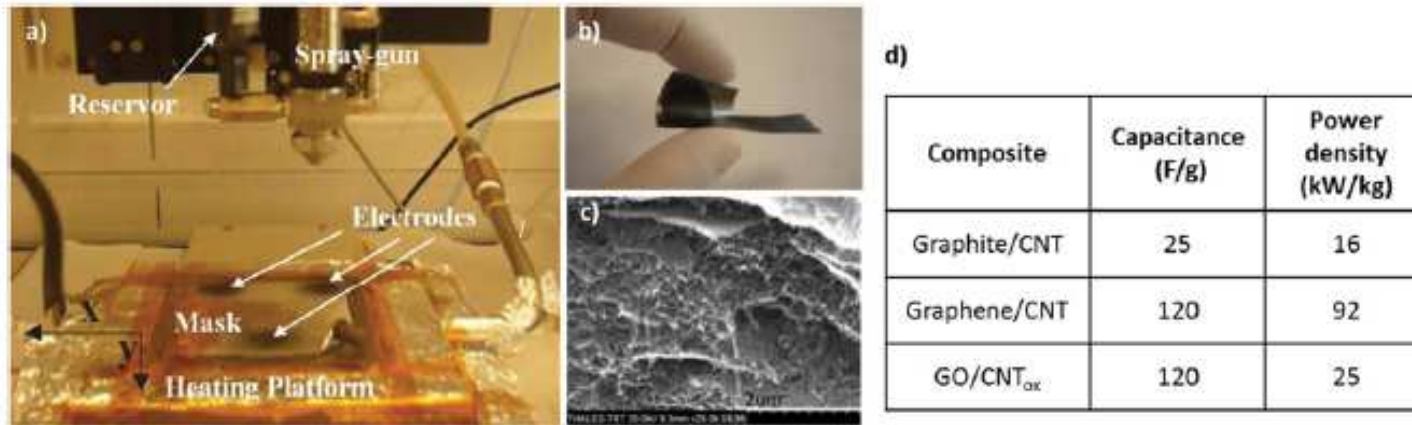


Figure 11. (a) Dynamic spray-gun deposition set-up used for the deposition of mixtures of nanomaterial graphite collectors; (b) flexible electrode deposited on a graphite collector; (c) SEM cross section of a graphite collector covered with a mixture of sprayed graphene/graphite (50%)/CNTs (50%); (d) typical electrode performance versus composite material.

Contracts and Objectives

GRAPHENE FLAGSHIP

➤ Results published in « 2D Materials » and presented at IEEE NANO 2015 ([link2](#))

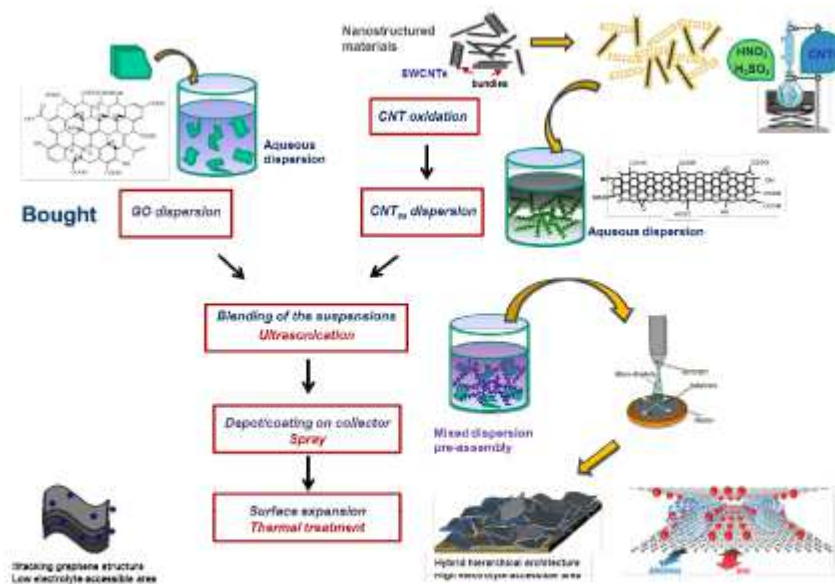


Fig3: Process to obtain stable suspensions of mixture of GO and oxidised Carbon Nanotubes (ox-CNTs).

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Graphene Oxide based electrodes

Advantages

- Water based suspensions
- Low cost material
- Very stable suspensions (months, years?)
- Capacitance of 120F/g, Power density of 30kW/Kg

Drawbacks

- Power lower than for Graphene (factor three)

Graphene based electrodes

Advantages

- Same capacitance than GO but Larger power density demonstrated (~100kW/Kg)

Drawbacks

- NMP based suspensions (toxic and higher boiling temperature than water)
- Stability of the suspensions (weeks?)



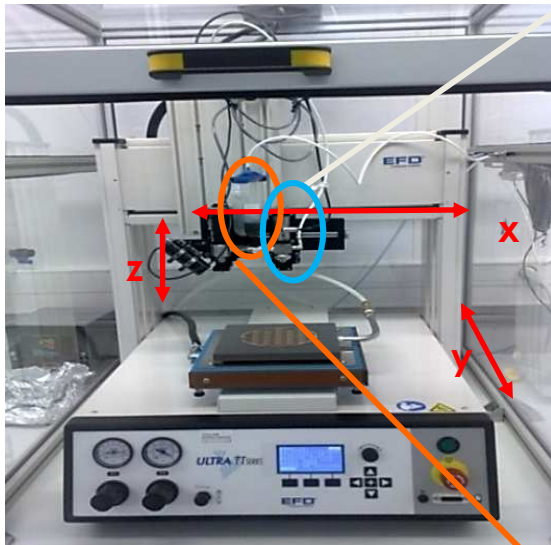
Video 1



- Last developments on nanostructuration

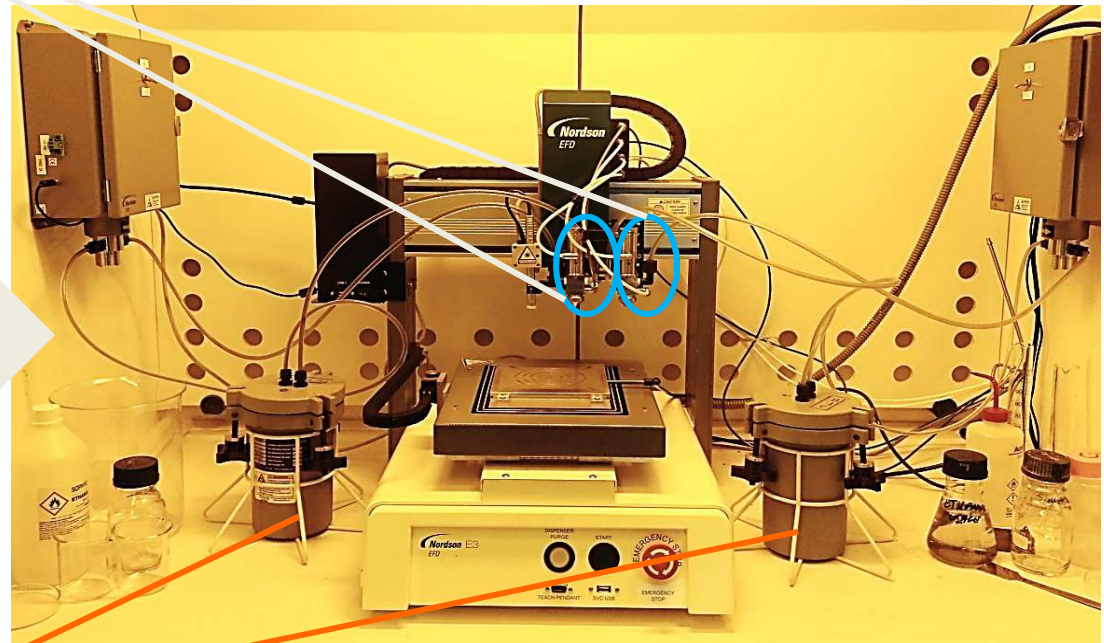
New machine with two nozzles acquired by Thales (form January 2017)

Nozzles

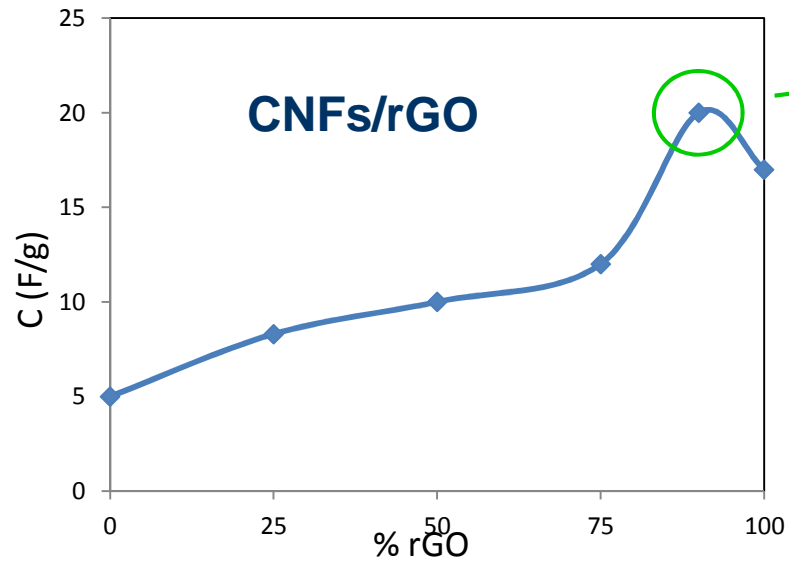


Only one nozzles (old machine)

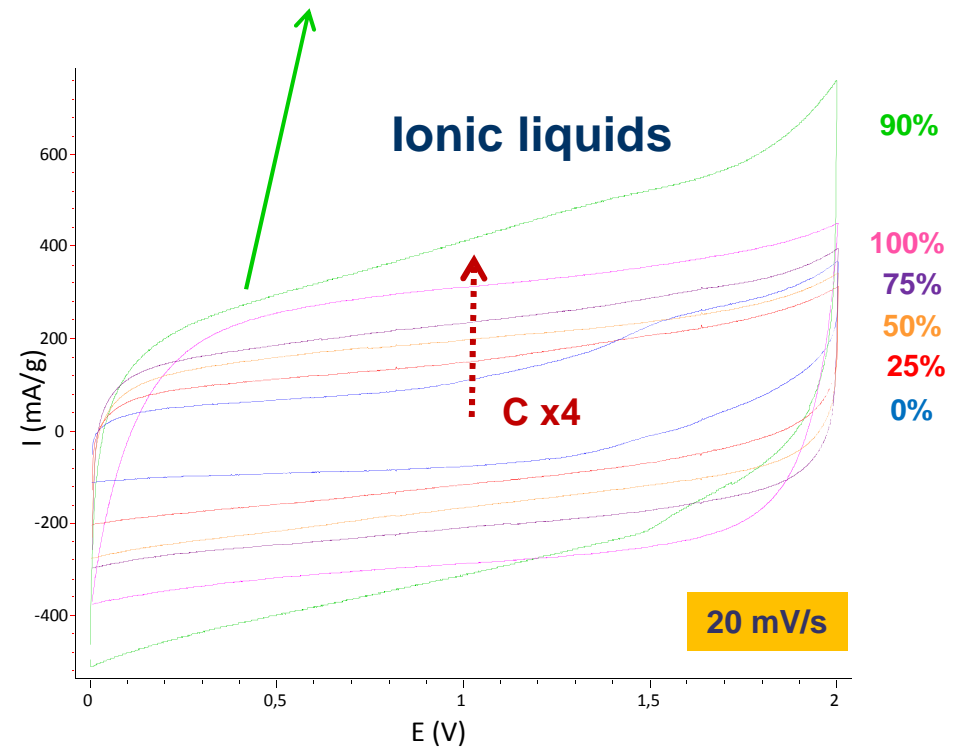
Reservoirs



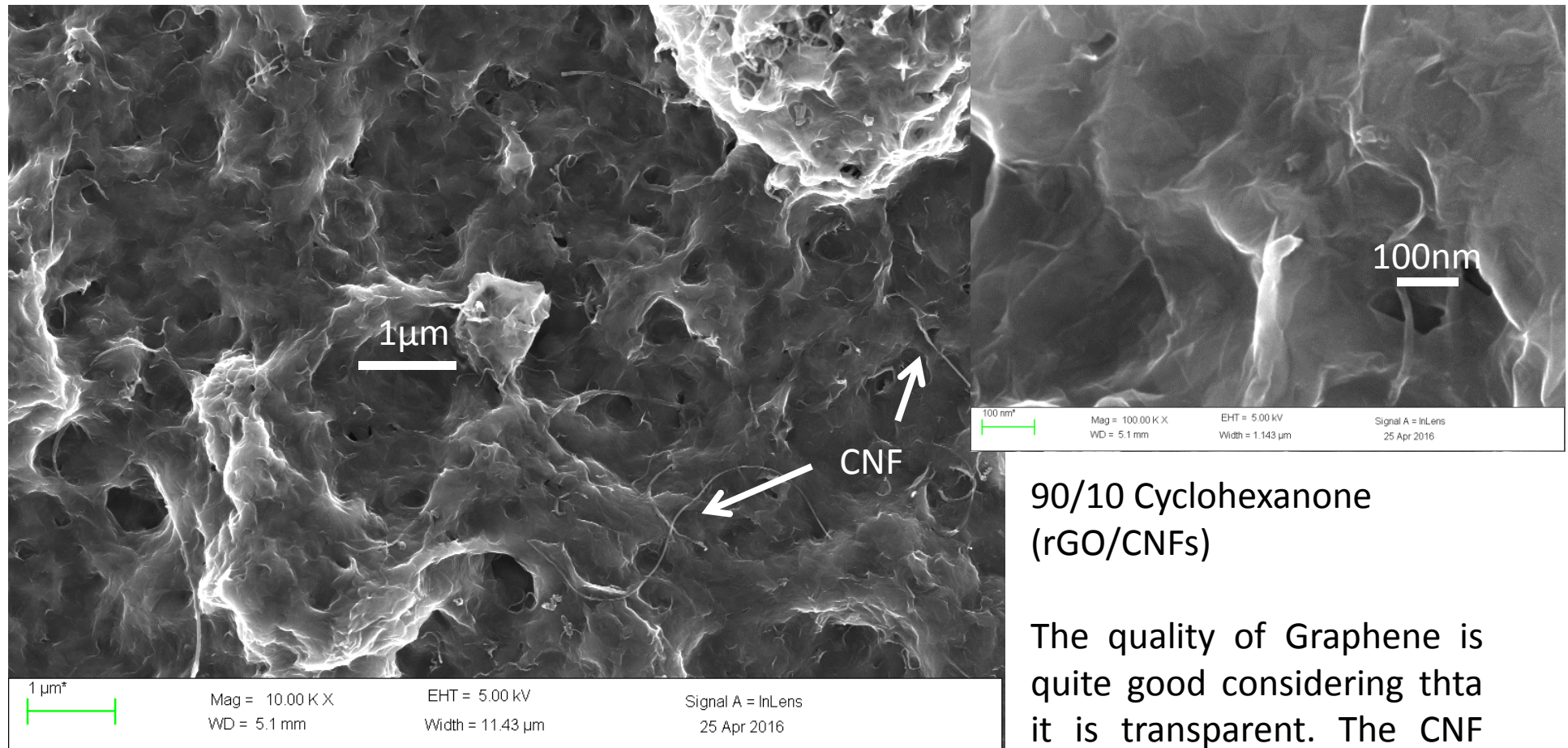
Utilisation of specific ionic liquids : large temperature interval for avionics (-55°C +105°C)



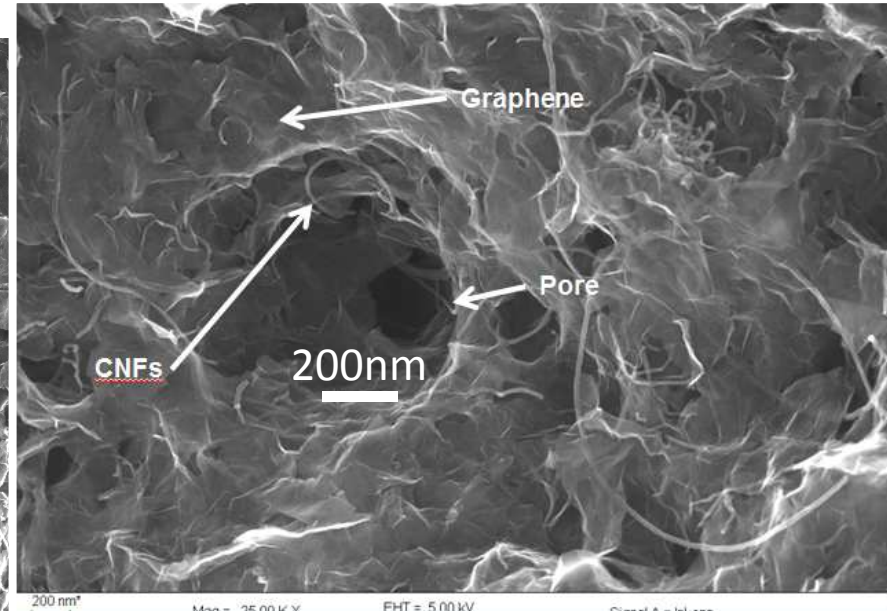
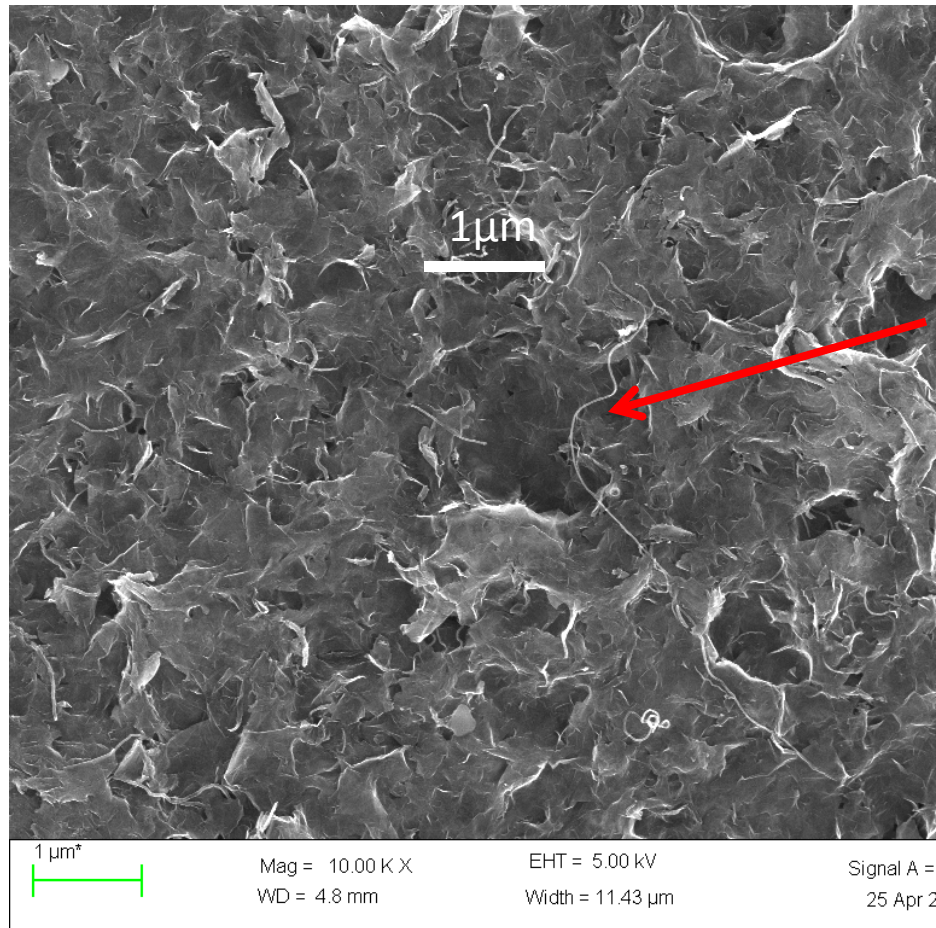
$C = 30 \text{ F/g}$ (whole cell)
 $P = 40 \text{ kW/Kg}$



Deelopments using different solvents



Deelopments using different solvents



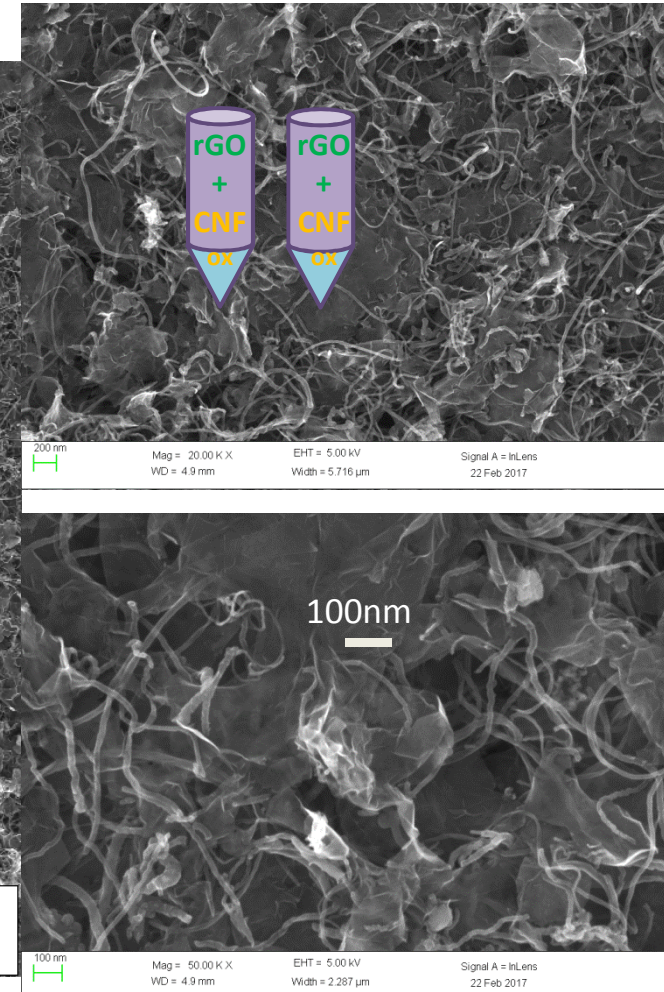
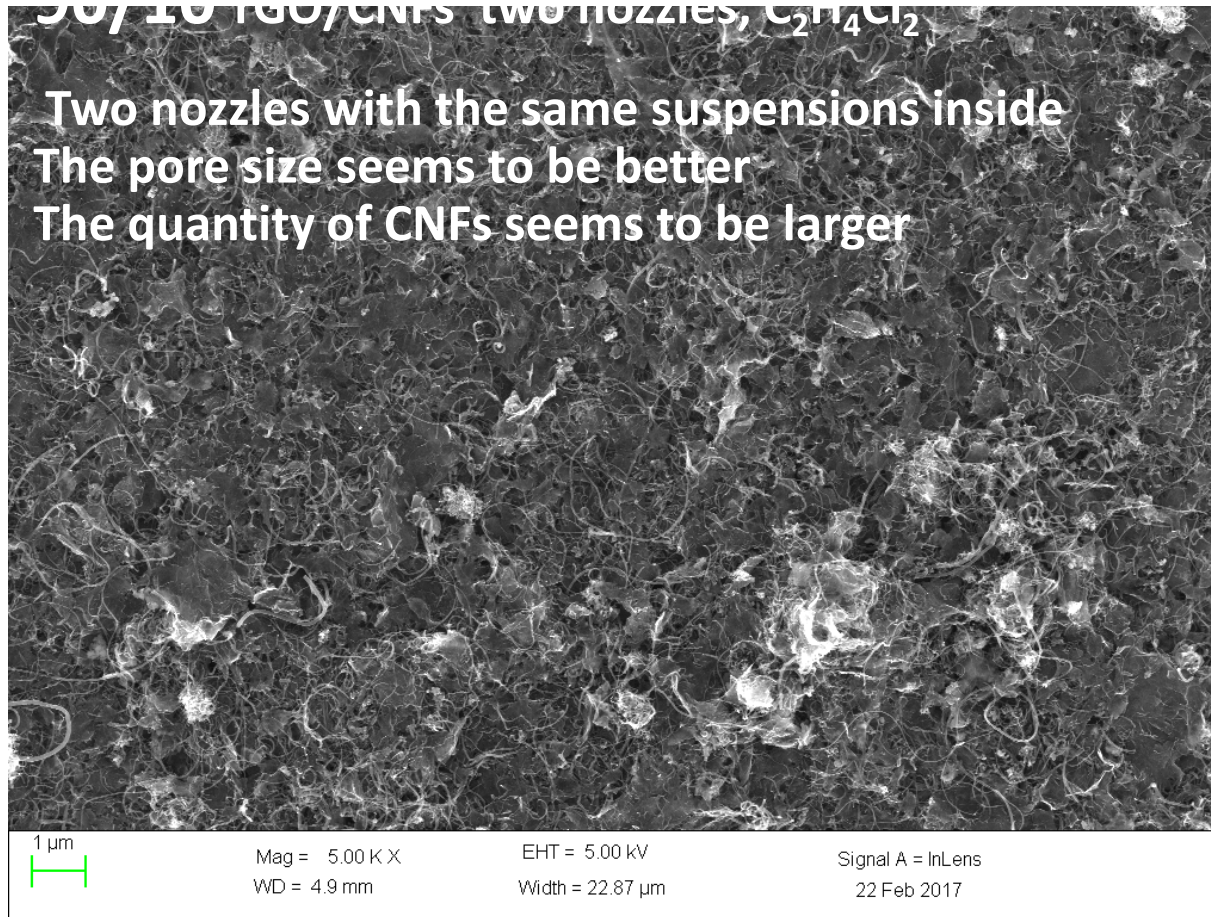
90/10 Cyclohexanone
(rGO/CNFs)

The quality of Graphene is quite good considering thta it is transparent. The CNF

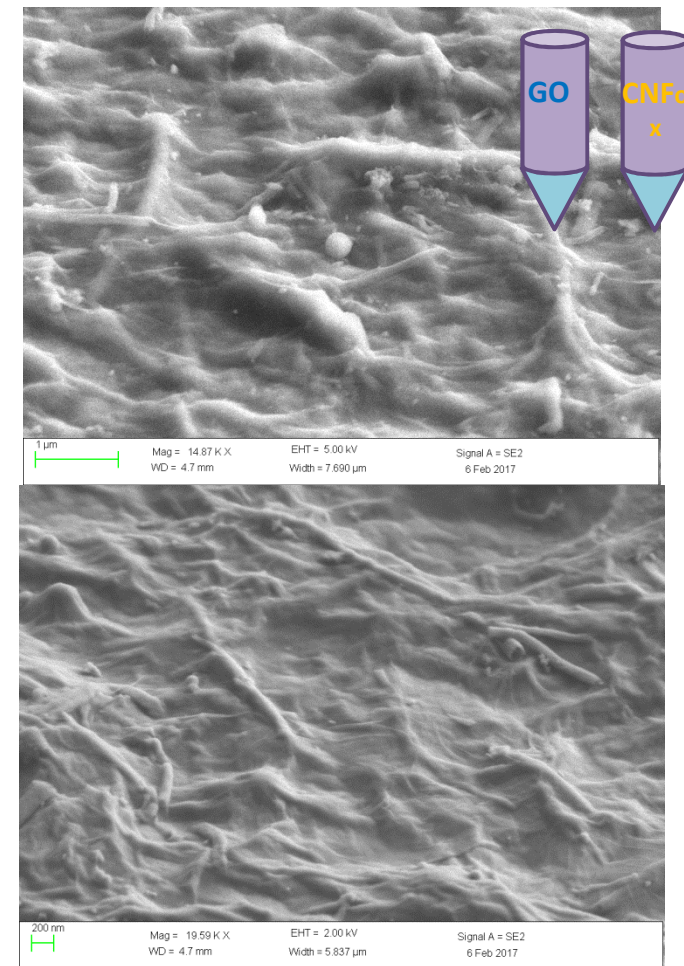
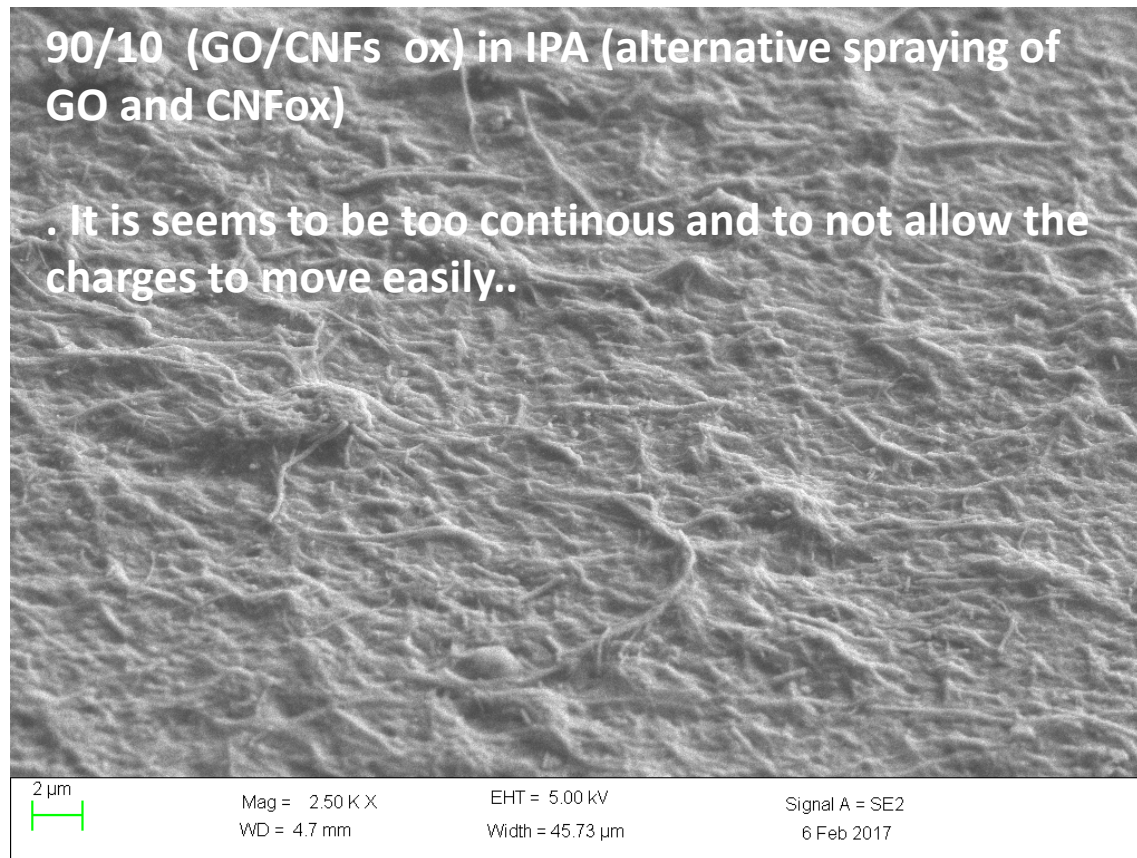
Deelopments using different solvents

50/10 rGO/CNFs two nozzles, $C_2H_4Cl_2$

Two nozzles with the same suspensions inside
The pore size seems to be better
The quantity of CNFs seems to be larger



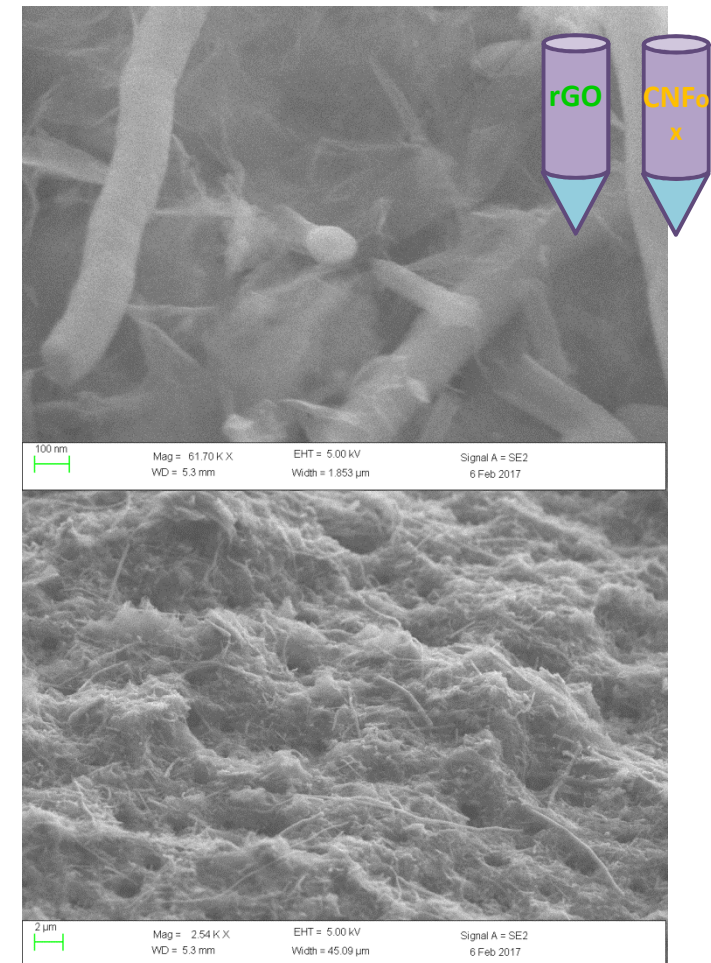
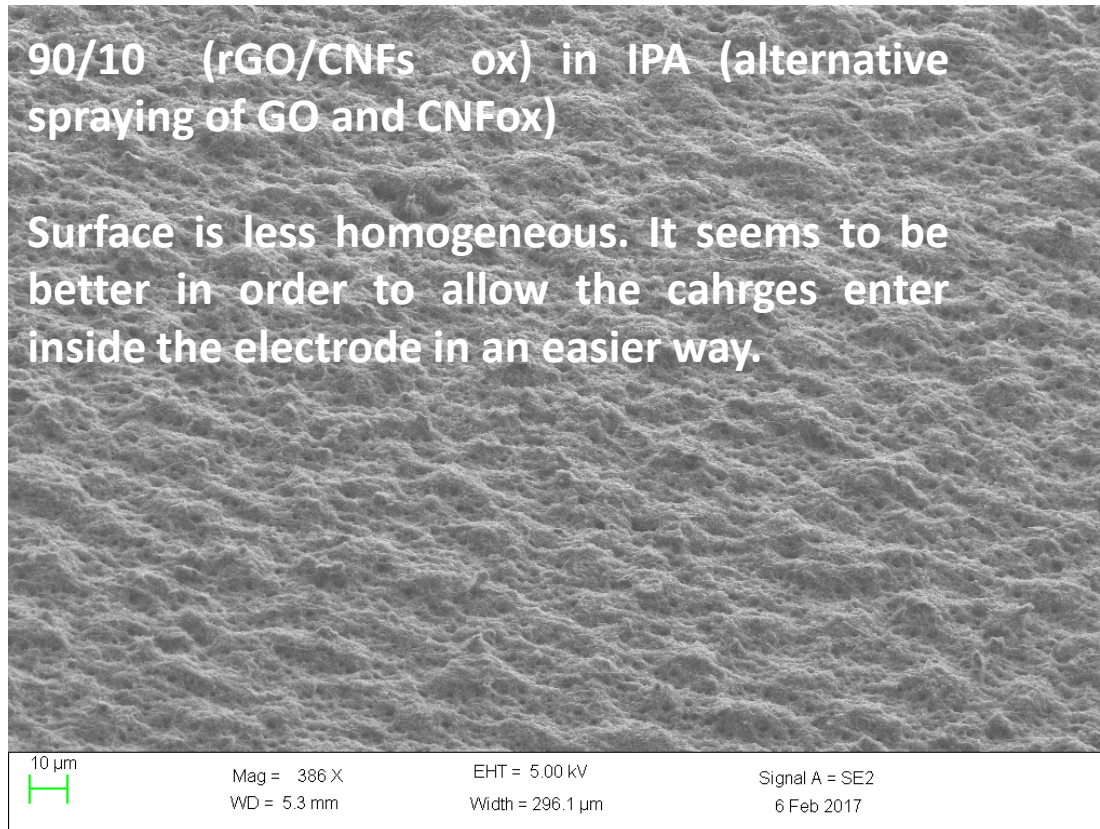
Developments using different solvents



Developments using different solvents

90/10 (rGO/CNFs ox) in IPA (alternative spraying of GO and CNFox)

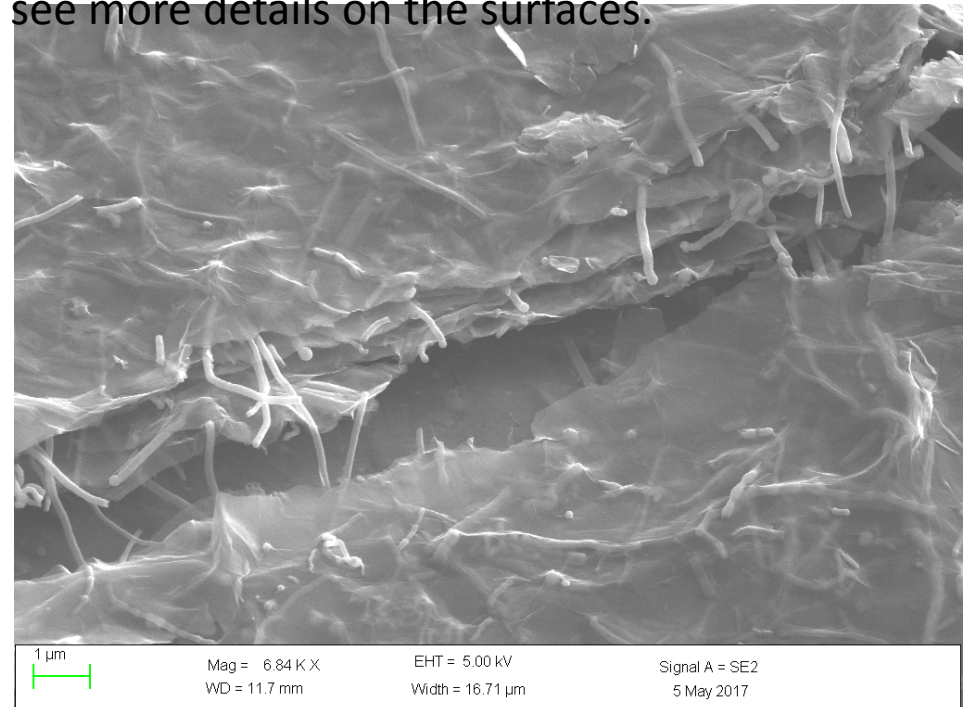
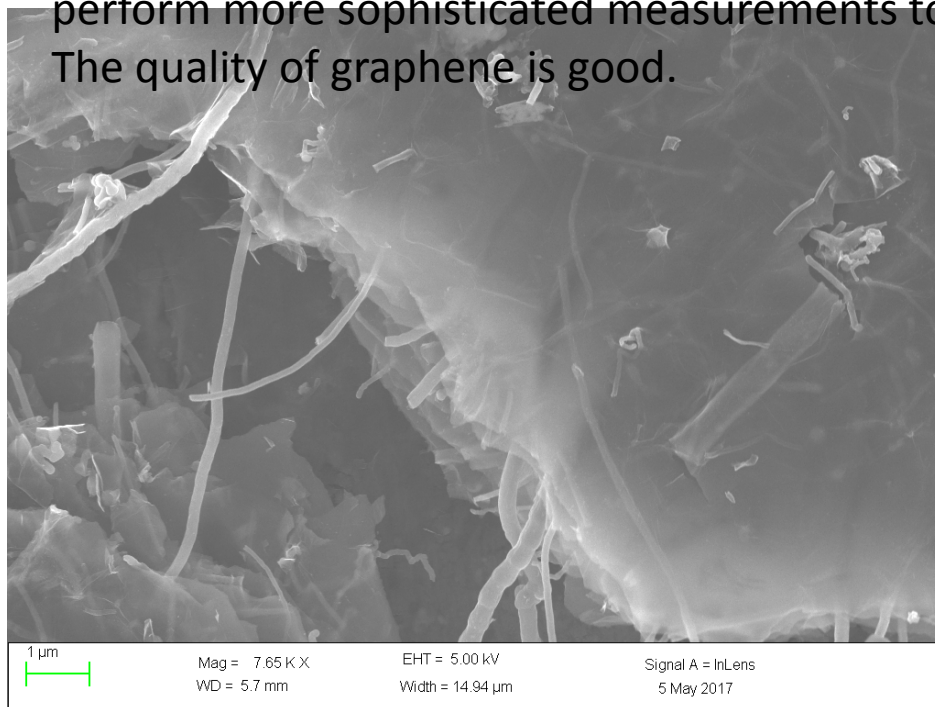
Surface is less homogeneous. It seems to be better in order to allow the cahrges enter inside the electrode in an easier way.



Developments using different solvents

75/25 Cyclohexanone (GO/CNFs ox) (alternative spraying of GO and CNFox)

The intercalation seems very good. The only issue is about the graphene layer. It seems to be too continuous and to not allow the charges to move easily. We have to perform more sophisticated measurements to see more details on the surfaces. The quality of graphene is good.



Fundings



GRAPHENE FLAGSHIP



Work Package 9: Energy

Work Package Leader - Dr. Etienne Quesnel, CEA French Alternative Energies and Atomic Energy Commission, France
Work Package Leader - Dr. Vittorio Pellegrini, Italian Institute of Technology, IIT graphene labs, Italy



The FIBRALSPEC project is supported by the
European Commission under the 7th Framework Programme
and will run for 48 months from January 2014 to December 2017.

Project coordinator: Costas A. Charitidis

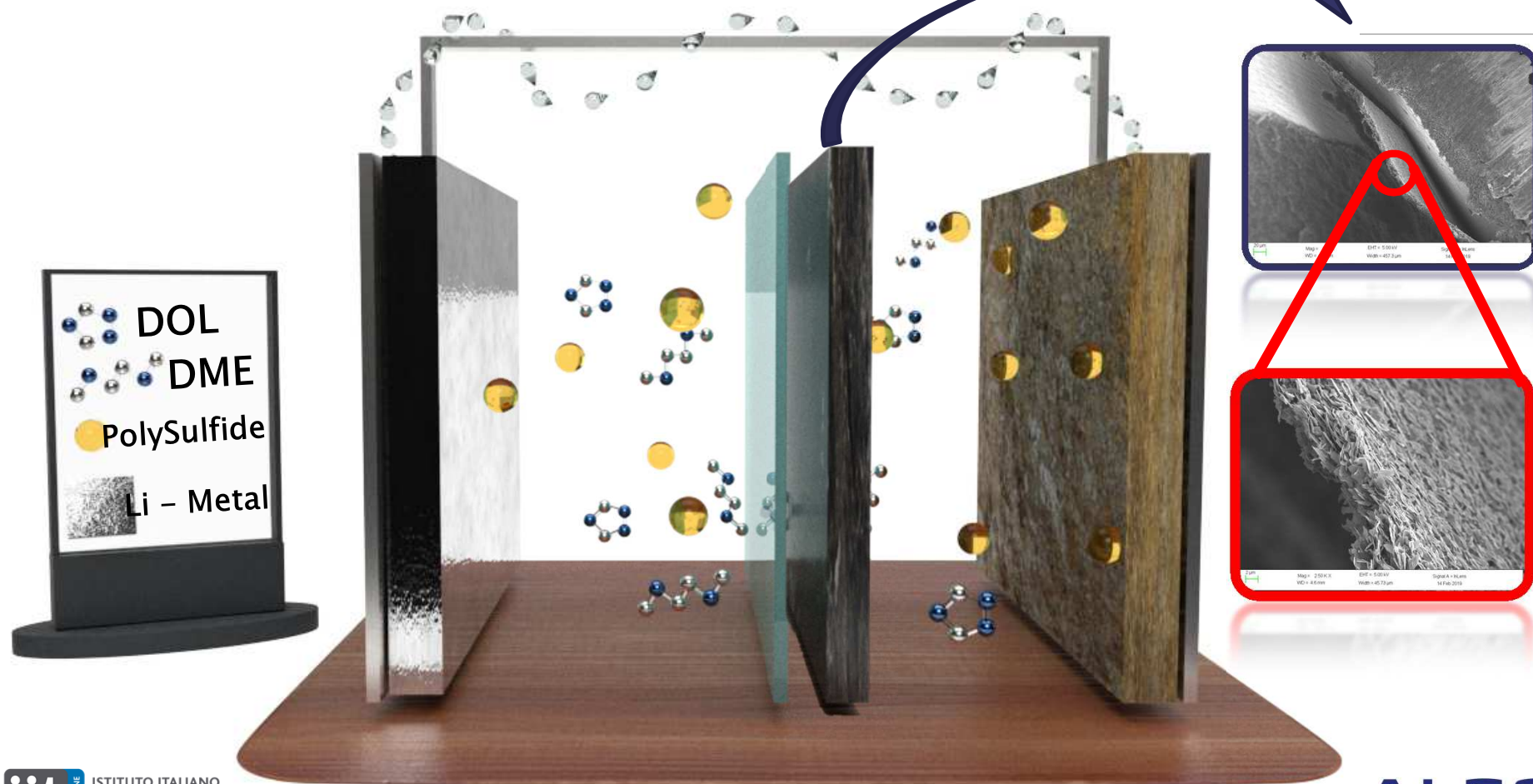


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Outline

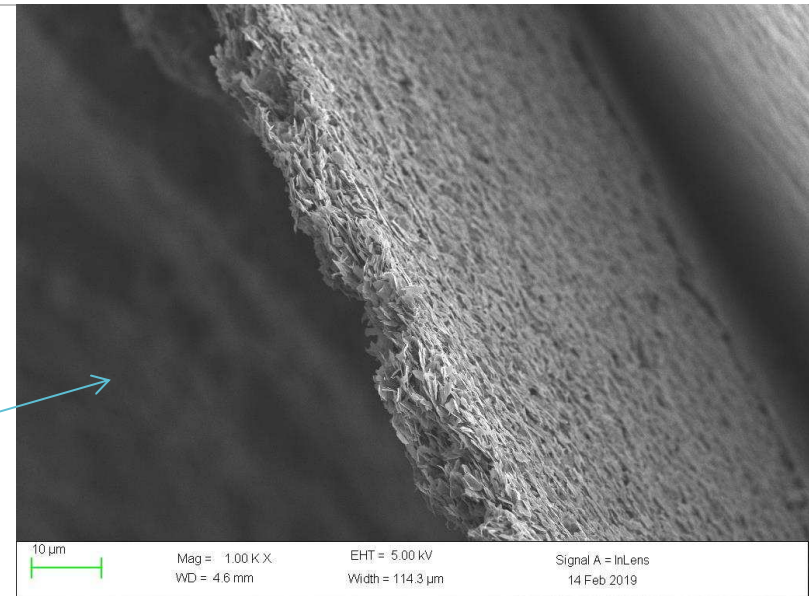
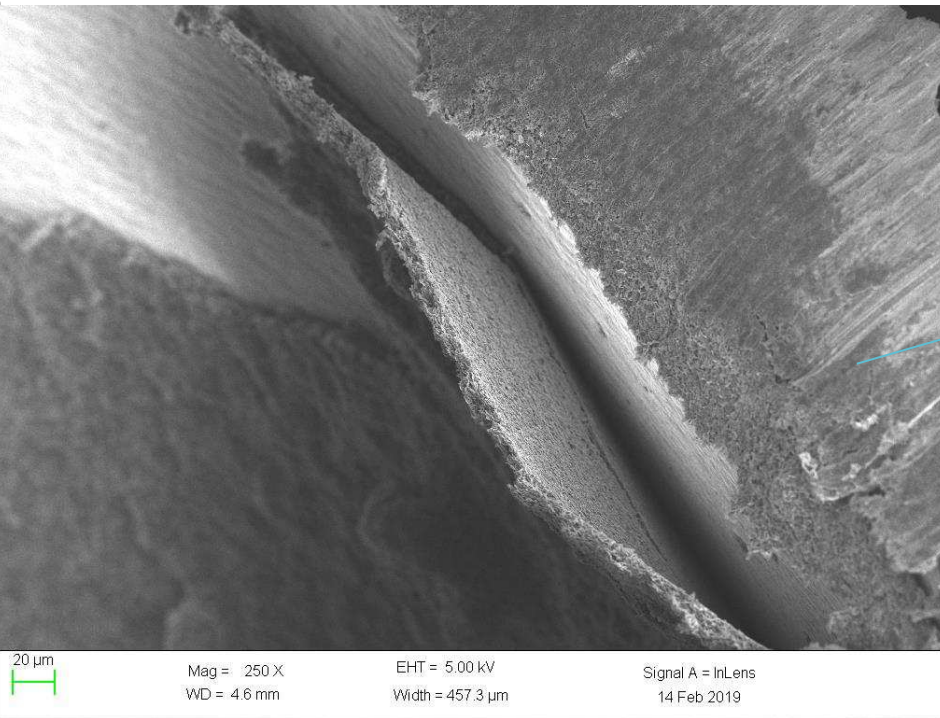
- Gas sensors application
- Supercapacitors based on mixtures of Graphene/CNTs (Energy)
- Graphene oxide based memories
- Electro Magnetic Shielding
- **Li-S batteries**

Carbon Interlayer to avoid the polysulfide shuttle effect

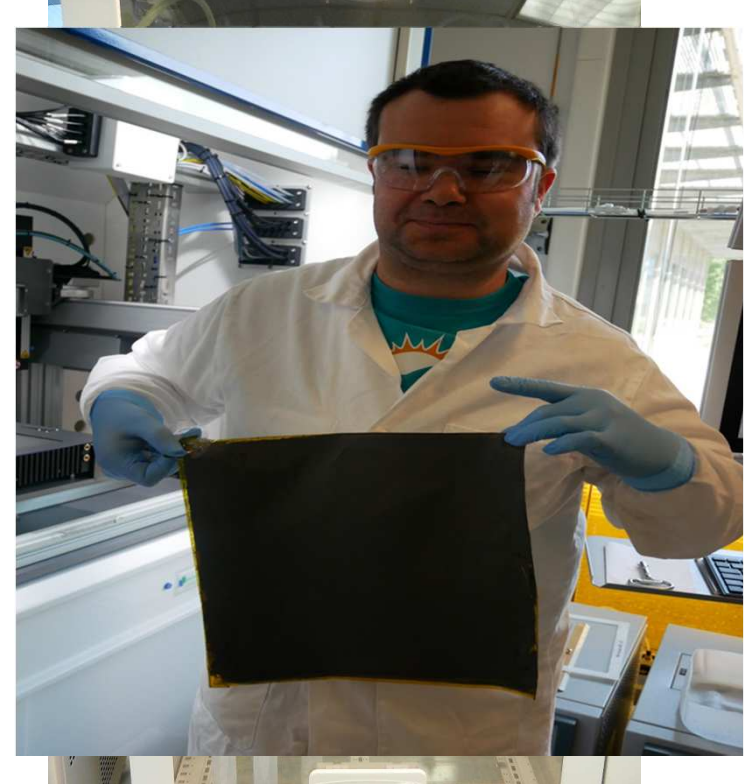
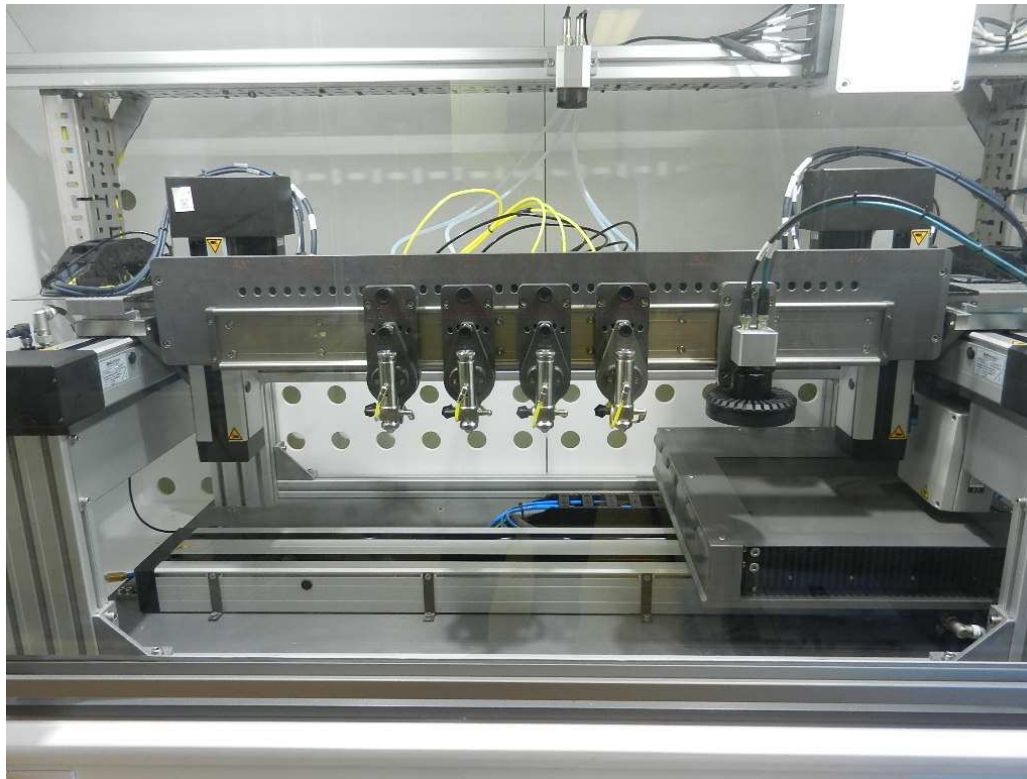


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New Machine funded by the Graphene Flagship

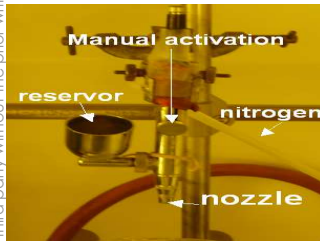


Conclusions

Spray-gun deposition method evolution

Deposition method evolution

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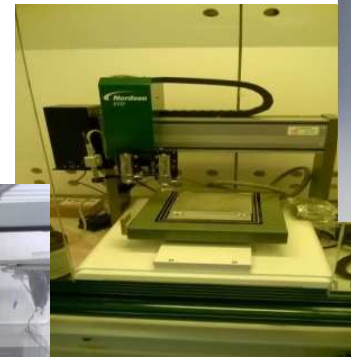
2006
Manual Air-brush
SMALL SURFACES
MANUAL SYSTEM



2007
Semi-automatic
Air-brush



2008
Spray-gun robot
with one nozzle



2016
Spray-gun robot
with two nozzles



2017
Spray-gun robot
with four nozzles
pre-industrialized
(M-Solv)

LARGE SURFACES
COMPLETELY AUTOMATISED
SYSTEM
(pre-industrial prototype)

Roll to Roll Core 2 (2020) Pilot Production line using Spray



Disclosed to Thales for Graphene Flagship



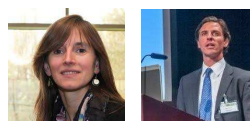
- IIT for exfoliated Graphene



- NTUA for carbon nanofibers



- Graphenea for GO in water



- M-Solv for collaboration in developing the new machine



Projet en cours qui concernent le spray



ICT FET Flagship Core 2 2018-2020



H2020 NMBP 2018-2022



H2020 NMBP 2015-2019



ANR France-Luxembourg 2019-2022

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IDTechEx Show!

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11 - 12 April 2018 | Estrel Convention Center, Berlin, Germany
Connecting Emerging Technologies With Global Brands



GRAPHENE WEEK 2018
 10-14 September, San Sebastian, Spain

GRAPHENE FLAGSHIP



2D MATERIALS FOR ENVIRONMENT AND ENERGY APPLICATIONS
 GOTHENBURG (HINDÅS), SWEDEN, 1-6 JULY 2018

GRAPHENE STUDY



Graphene
 2018
 June 26 - 29
 Dresden (Germany)

2018
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 Bringing together
 Nanoscience & Nanotechnology


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 BARCELONA 26 FEB-1 MAR 2018

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PRINTED ELECTRONICS
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BARCELONA, SPAIN

14-15 JUNE | 2018

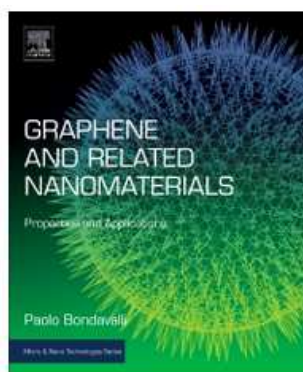


2018 **MRS**
 FALL MEETING & EXHIBIT
 November 25-30, 2018 | Boston, Massachusetts

2018 **MRS**
 SPRING MEETING & EXHIBIT
 April 2-6, 2018 | Phoenix, Arizona

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- **High-power graphene-carbon nanotube hybrid supercapacitors.** ChemNanoMat. Accepted Author Manuscript. Ansaldo, A., Bondavalli, P., Bellani, S., Del Rio Castillo, A. E., Prato, M., Pellegrini, V., Pognon, G. and Bonaccorso, F. (2017), doi:10.1002/cnma.201700093
- **Graphene-based technologies for energy applications, challenges and perspectives,** Etienne Quesnel, Frédéric Roux, **Paolo Bondavalli** et al., *2D Materials* 08/2015; 2(3):030204.
- **Supercapacitor electrode based on mixtures of graphite and carbon nanotubes deposited using a new dynamic air-brush deposition technique,** P Bondavalli, C.Delfaure, P.Legagneux, D.Pribat JECS 160 (4) A1-A6, **2013**
- **Non-faradic carbon nanotubes based supercapacitors : state of the art,** P.Bondavalli, D.Pribat, C.Delfaure, P.Legagneux, L.Baron, L.Gorintin, J-P. Schnell, Eur. Phys. J. Appl. Phys. 60,10401, **2012**
- **Graphene based supercapacitors fabricated using a new dynamic spray-gun deposition technique,** P. Bondavalli and G. Pognon, 2015 IEEE 15th International Conference on Nanotechnology (IEEE-NANO), Rome, 2015, pp. 564-567.
- Graphene and Related Nanomaterials 1st Edition, Properties and Applications, **Paolo Bondavalli**, Hardcover ISBN: 9780323481014, Elsevier, Published Date: 20th October September 2017



Graphene and Related Nanomaterials

1st Edition

Properties and Applications

Authors: Paolo Bondavalli

THALES

- WO/2016/124756 (A1) - **Method of depositing oxidized carbon-based microparticles and nanoparticles**, Pognon Grégory [fr]; Bondavalli Paolo [fr]; Galindo Christophe [fr]
- WO/2014/166952 (A1) - **Electrode-gel electrolyte assembly comprising a porous carbon material and obtained by radical polymerisation**, Le Barny, Pierre; (FR). Divay, Laurent; (FR). Galindo, Christophe; (FR)
- FR2988900 (A1) - **Electrode pour supercondensateur**, Christophe Galindo, Laurent Divay, Le Barny Pierre
- FR2989215 (A1) - **Electrode hybride pour supercondensateur**, , Le Barny Pierre, Laurent Divay, Christophe Galindo
- FR2989821 (A1) - **Electrode hybride nanotubes de carbone-carbone mesoporeux pour supercondensateur**, Le Barny Pierre, Laurent Divay, Christophe Galindo
- FR3011671 (A1) - **Collecteur de courant pour supercapacite**, Bondavalli Paolo [fr]; Legagneux Pierre [fr]; Galindo Christophe
- WO2012049428 (A2) - **Method for depositing nanoparticles on a surface and corresponding nanoparticle depositing appliance**, Bondavalli Paolo [fr]; Gorintin Louis [fr]; Legagneux Pierre [fr]; Ponard Pascal [fr]
- FR2976118 (A1) - **Method for manufacturing collector-electrode assembly that is utilized in supercapacitor, involves forming collector and electrode by spraying suspension comprising nano/microparticles suspended in liquid in substrate**, Bondavalli Paolo [fr]; Schnell Jean Philippe [fr]; Legagneux Pierre [fr]; Gorintin Louis [fr]
- EP2769395 (A1) - **Collector-electrode assembly which can be integrated into an electrical energy storage device**, Legagneux Pierre [fr]; Bergonzo Philippe [fr]; Bondavalli Paolo [fr]; Mazellier Jean-Paul [fr]; Scorsone Emmanuel [fr]

Thank you for your attention!!!

